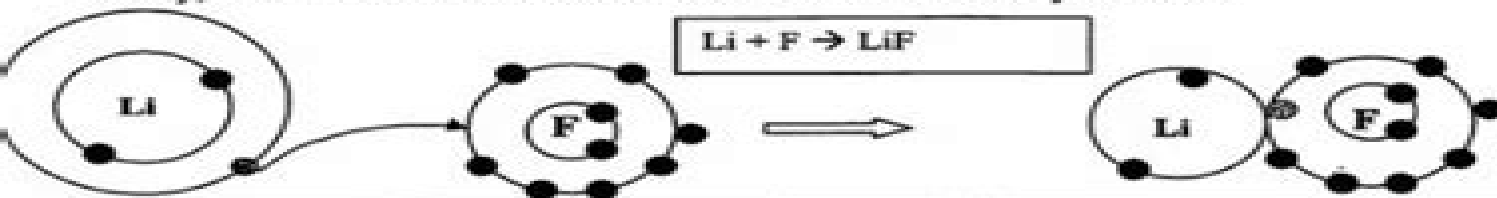


### Ionic Bonding Worksheet

For each pair of elements below draw an atomic diagram showing electrons in different energy levels. Draw arrows to show where the outer electrons will go during a chemical reaction, then draw the resulting compound. Finally, fill in the table below each reaction. Refer to the sample shown.



Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
Li	1	1 to F	$Li^+$
F	7	1 from Li	$F^-$

Reactions	Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
1) $Li + Cl \Rightarrow LiCl$ 	Li	1	1 to Cl	$Li^+$
	Cl	7	1 from Li	$Cl^-$
2) $Ca + O \Rightarrow CaO$ 	Ca	2	2 to O	$Ca^{2+}$
	O	6	2 from Ca	$O^{2-}$
3) $Be + F \Rightarrow BeF_2$ 	Be	2	1 to each F	$Be^{2+}$
	F	7	Each 1 gets 1 from Be	$F^-$
4) $Mg + S \Rightarrow MgS$ 	Mg	2	2 to S	$Mg^{2+}$
	S	6	2 from Mg	$S^{2-}$
5) $K + F \Rightarrow KF$ 	K	1	1 to F	$K^+$
	F	7	1 from K	$F^-$

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