Reference Tables for Physical Setting/EARTH SCIENCE

Radioactive Decay Data

RADIOACTIVE ISOTOPE	DISINTEGRATION	HALF-LIFE (years)
Carbon-14	14°C• 14N	5.7×10^3
Potassium-40	*OX -*OAr	1.3 × 10 ⁹
Uranium-238	²³⁶ U → ²⁰⁶ Pb	4.5 × 10 ⁹
Rubidium-87	⁶⁷ Rb — ⁸⁷ Sr	4.9 × 10 ¹⁰

Equations

Eccentricity = di	stance between foci ength of major axis
Gradient = chang	ge in field value distance
Rate of change =	change in value time
Density = $\frac{\text{mass}}{\text{volume}}$	

Specific Heats of Common Materials

MATERIAL	SPECIFIC HEAT (Joules/gram • °C)		
Liquid water	4.18		
Solid water (ice)	2.11		
Water vapor	2.00		
Dry air	1.01		
Basalt	0.84		
Granite	0.79		
lirom	0.45		
Copper	0.38		
Lead	0.13		

Properties of Water

Heat	energy	gained during melting 334 J/g
Heat	energy	released during freezing 334 J/g
Heat	energy	gained during vaporization 2260 J/g
Heat	energy	released during condensation 2260 J/g

Average Chemical Composition of Earth's Crust, Hydrosphere, and Troposphere

ELEMENT (symbol)	CRUST		HYDROSPHERE	TROPOSPHERE
	Percent by mass	Percent by volume	Percent by volume	Percent by volume
Oxygen (O)	46.10	94.04	33.0	21.0
Sillicon (Si)	28.20	0.88		
Aluminum (Al)	8.23	0.48		
Iron (Fe)	5.63	0.49		
Calcium (Ca)	4.15	1.18		
Sodium (Na)	2.36	1,11		
Magnesium (Mg)	2.33	0.33		
Potassium (K)	2.09	1.42		1
Nitrogen (N)				78.0
Hydrogen (H)			66.0	
Other	0.91	0.07	1.0	1.0

2011 EDITION

This edition of the Earth Science Reference Tables should be used in the classroom beginning in the 2011–12 school year. The first examination for which these tables will be used is the January 2012 Regents Examination in Physical Setting/Earth Science.

Eurypterus remipes

. New York State Fossil

20topical Earth Science Reference Tables Review

Thomas McGuire

20topical Earth Science Reference Tables Review:

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