

Match the notation below with the atom described in questions 1-4:



- 1) Represents an atom that is in an excited state
- 2) Represents an atom that is a noble gas
- 3) Represents an atom that is a transition metal
- 4) Represents an atom of an alkali earth metal

Questions 5-8 refer to the following descriptions of bonding in different types of solids.

- a) Lattice of positive and negative ions held together by electrostatic forces
  - b) Closely packed lattice with delocalized electrons throughout giving ability to conduct electricity and permitting ductility
  - c) Strong single covalent bonds making up a molecule with a low molecular mass
  - d) Double and/or triple bonds are used to hold atoms together
  - e) Expanded octet of electrons on the central atom
- 5) Potassium Chloride,  $\text{KCl}_{(s)}$
  - 6) Gold,  $\text{Au}_{(s)}$
  - 7) Methane,  $\text{CH}_{4(s)}$
  - 8) Sulfur Hexafluoride,  $\text{SF}_{6(s)}$

Questions 9-12 refer to the reactions represented below. Answers may be used more than once.



- 9) A reaction with both a weak and a strong acid as reactants
- 10) A precipitation reaction in which  $\text{Ca}^{2+}$  could be a spectator ion
- 11) A reaction in which the same reactant undergoes both oxidation and reduction
- 12) A combustion reaction

# 20multiple Choice Exam Ap Chemistry

**Padhraic Smyth**



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