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Chemistry: Supplemental Stoichiometry Problems

Directions: Solve each of the following problems. Assume excess of any reactant that isn't mentioned, unless otherwise specified. Show your work, including proper units, to earn full credit.

- 1. Given the balanced equation, show what the following molar ratios (, the conversion factors) should be. Include units in all terms 2 C_aH₁₀ + 13 O₂→ 8 CO₂ + 10 H₂O
- a. C₄H₁₀ / O₂ b. O₂ / CO₂c. O₂ / H₂O d. C₄H₁₀ / CO₂ e. C₄H₁₀ / H₂O
- How many moles of oxygen are made if 12.0 moles of potassium chlorate react? 2 KClO₃→ 2 KCl + 3 O₃

Q3 involves the reaction: Copper(II) chloride reacts w/sodium nitrate to produce copper(II) nitrate and sodium chloride.

- 3A. Write the balanced equation for the reaction.
- 3B. If 20.0 g of copper(II) chloride react with 20.0 g of sodium nitrate, what mass of sodium chloride is formed?

- 3C. What is the limiting reactant? _____
- 3D. How many moles of copper(II) nitrate are formed?
- 3E. What mass of excess reactant is left over?

ANSWERS: 2. 18.0 mol O₂ 38. 13.8 g NaCl 3C. NaNO₂ 3D. 0.118 mol Cu(NO₂)₂ 3E. 4.2 g CuCl₂

1 Century Chemistry Supplementary 2b Answer

John L Burmeister, Fred Basolo

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chemistry This is the case in the currently significant areas of research including organometallic homogenous catalysis biological reactions of metalloproteins and even the solid state extended structures of new materials In these systems the metals are of primary importance because they are the sites of ligand substitution or redox reactions. In the solid materials the coordination number of the metal and its stereochemistry are of major importance Some fifty years of research on transition metal complexes carried out in the laboratory of Professor Basolo at Northwestern University is recorded here as selected scientific publications The book is divided into three different major research areas each dealing with some aspect of coordination chemistry. In each case introductory remarks are presented which indicate what prompted the research projects and what the major accomplishments were Although the research was of the academic curiosity driven type some aspects have proven to be useful to others involved in projects that were much more applied in nature The Educational Times, and Journal of the College of Preceptors, 1916 The Publishers' Trade List Annual ,1904 The World Review ,1927 The United States Catalog ,1928 The United States Catalog George Flavel Danforth, Marion Effie Popular Educator ,1895 Rochdale free public library. Supplementary Potter, 1900 The Academy ,1899 catalogue of books in the lending department George Hanson (librarian.), 1889

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