

Name: _____ Period: _____ Date: _____



Acid Base Review Sheet (chapter 19)



VOCABULARY: Flashcards must be made for the following words and be handed in with the work below on the day of the test or you will not receive any credit for the review sheet.

- | | | |
|-------------------|---------------------|---------------|
| • Arrhenius Acid | • Polyprotic acid | • pOH |
| • Arrhenius Base | • Acidic Solution | • salt |
| • Bronsted Acid | • Conjugate acid | • strong acid |
| • Bronsted Base | • Conjugate Base | • strong base |
| • Indicator | • End point | • weak acid |
| • Amphoteric | • Equivalence point | • weak base |
| • Basic solution | • Titration | |
| • Monoprotic acid | • pH | |

Acid-Base Theory

1. Fill in the chart below with the different views of acids and bases according to the different scientists.

	Arrhenius	Bronsted-Lowry
Acids		
Bases		

2. Identify the acid, base, conjugate acid, and conjugate base in the following reactions.



Base Acid CA CB

c) Whose theory are conjugate acids and bases dependent upon? _____

3. Label the following as an Arrhenius acid or Arrhenius base or neither.



4. a) List properties of acids: _____

b) List properties of bases: _____

5. Identify the common substances below as being acidic, basic, or neutral:

• lemon juice _____

• coffee _____

• grapefruit _____

• milk _____

• water _____

• soap _____

• soda _____

• banana _____

6. Identify the following acids as monoprotic or polyprotic. If the acid is polyprotic, also label it as either diprotic or triprotic.



6. Using the diagram below, explain why water is considered to be amphoteric.



19 Acids And Bases Reviewsheet Answers

Neil E. Schore



19 Acids And Bases Reviewsheet Answers:

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