

Polyatomic Ions Worksheet Answer Key

Polyatomic Practice

1. Name or write the formula for the following polyatomic ions

sulfate	<u>SO_4^{2-}</u>	<u>CO_3^{2-}</u>	carbonate
nitrite	<u>NO_2^-</u>	<u>MnO_4^{1-}</u>	manganate
perphosphate	<u>PO_4^{3-}</u>	<u>SO_5^{2-}</u>	persulfate
hypoiodite	<u>IO^-</u>	<u>BrO_2^{1-}</u>	bromite
chlorite	<u>ClO_2^-</u>	<u>CO_4^{2-}</u>	percarbonate
phosphite	<u>PO_3^{3-}</u>	<u>PO_5^{3-}</u>	perphosphate
percarbonate	<u>CO_4^{2-}</u>	<u>ClO^{1-}</u>	hypochlorite
bromate	<u>BrO_3^-</u>	<u>IO_2^{1-}</u>	iodite
hyposulfite	<u>SO_2^{2-}</u>	<u>PO_4^{3-}</u>	phosphate
permanganate	<u>MnO_4^-</u>	<u>NO_2^{1-}</u>	nitrite
carbonite	<u>CO_2^{2-}</u>	<u>SO_4^{2-}</u>	sulfate

2. Name or write the formula for the following Type I polyatomic ionic compounds

beryllium hydroxide	<u>$\text{Be}(\text{OH})_2$</u>	<u>$\text{Ba}(\text{IO}_3)_2$</u>	barium iodate
sodium nitrite	<u>NaNO_2</u>	<u>$\text{Ga}(\text{CNO})_3$</u>	gallium cyanate
ammonium chloride	<u>NH_4Cl</u>	<u>Ag_2SO_3</u>	silver sulfite
calcium bisulfate	<u>$\text{Ca}(\text{HSO}_4)_2$</u>	<u>MgCO</u>	magnesium hypocarbonite
rubidium perchlorate	<u>RbClO_4</u>	<u>NH_3NO_2</u>	ammonium nitrite
strontium sulfite	<u>SrSO_3</u>	<u>$\text{Al}(\text{C}_2\text{H}_3\text{O}_2)_3$</u>	aluminum acetate
aluminum acetate	<u>$\text{Al}(\text{C}_2\text{H}_3\text{O}_2)_3$</u>	<u>SrSO_5</u>	strontium persulfate
ammonium nitrate	<u>NH_4NO_3</u>	<u>RbClO_2</u>	rubidium chlorite
magnesium hypocarbonite	<u>MgCO</u>	<u>$\text{Ca}(\text{HSO}_4)_2$</u>	calcium bisulfate
silver hyposulfite	<u>Ag_2SO_2</u>	<u>NH_4Cl</u>	ammonium chloride

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Princeton Review



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Introductory Chemistry Charles H. Corwin,2011 To help students learn chemical skills and concepts more effectively Introductory Chemistry Concepts and Critical Thinking Sixth Edition highlights the connection between key concepts and key problem solving skills through critical thinking Math and problem solving are covered early in the text Corwin builds your problem solving ability through innovative learning aids and technology formulated to meet your needs This revision retains all the strengths of the previous editions while adding emphasis on conceptual understanding and critical thinking **Physics Briefs** ,1993 Investigating Chemistry Matthew Johll,2006-03-17 Matthew Johll s book introduces students from a non science background to the fundamentals of chemistry through an array of examples and applications from real life crime scenes Sherlock Holmes stories and authentic accounts of drug deals murders and thefts

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