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## The Atom and Nuclear Chemistry Review ANSWERS

1. Neutrons are neutral, but protons are positive and electrons are negative.
2. a) Francium-212                      How many neutrons? 125    How many protons? 87  
b)  $^{64}_{27}\text{Co}$                       How many neutrons? 37    How many protons? 27  
c) Give the **symbol** and **hyphenated name** for an atom with 20 protons and 22 neutrons  
 $^{42}_{20}\text{Ca}$       Calcium-42
3. What was the first element found to be radioactive? uranium
4. What is happening when an element is radioactive?  
The atom is spitting out particles from its nucleus.
5. What role do the neutrons play in keeping the nucleus of an atom stable? The stop the protons in the nucleus from "fighting" (i.e. from repelling each other)
6. Atoms with more than 83 protons in their nucleus are always radioactive.
7. Define **background radiation**. The radiation we receive from both natural and human-made sources.
8. What are some sources of naturally occurring radiation?  
Rocks, space, food, air
9. What are some human-made sources of radiation?  
x-rays, TV's, nuclear power plants, nuclear fallout from nuclear weapons testing, nuclear medicine
10. What is an alpha particle? A helium-4 atom  
  
Write its symbol.  $^4_2\text{He}$
11. What is a beta particle? A fast moving electron  
  
Write its symbol.  $^0_{-1}\text{e}$
12. Which is more harmful, an alpha particle or a beta particle? Why? How can each be stopped?  
An alpha particle can be stopped by a piece of paper, but to stop a beta particle something thicker like a book is needed. Because a beta particle is more penetrating, the beta particle is more dangerous.
13. Where do alpha and beta particles come from? the nucleus
14. When a beta particle forms, a neutron turns into a proton.

# 21 Review Modern Chemistry Answers Nuclear Chemistry

**Yijin Wang**



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