

NAME: _____

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AP CHEMISTRY MULTIPLE CHOICE QUESTIONS

Unit 8 – Introduction to Equilibrium



The equilibrium constant for the gas phase reaction above is 95 at 25°C. What is the value of the equilibrium constant for the following reaction at 25°C?



Reversed
+ Doubled

(A) 95²

(B) 95

(C) (95)^{1/2}(D) $\frac{1}{95}$ (E) $\frac{1}{(95)^2}$ 

$C_2H_4(g)$ is reduced by $H_2(g)$ in the presence of a solid platinum catalyst, as represented by the equation above. Factors that could affect the rate of the reaction include which of the following?

- I. Changes in the partial pressure of $H_2(g)$
- II. Changes in the particle size of the platinum catalyst
- III. Changes in the temperature of the reaction system

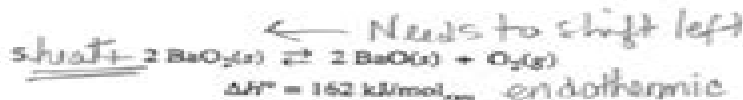
(A) III only

(B) I and II only

(C) I and III only

(D) II and III only

(E) I, II, and III



A sealed rigid vessel contains $BaO_2(s)$ in equilibrium with $BaO(s)$ and $O_2(g)$ as represented by the equation above. Which of the following changes will increase the amount of $BaO_2(s)$ in the vessel?

(A) Removing a small amount of $O_2(g)$ $R \times$ (B) Removing a small amount of $BaO(s)$ X (C) Adding He gas to the vessel X (D) Lowering the temperature L

2. For which of the following reaction mixtures at equilibrium would reducing the volume of the container at constant temperature cause the equilibrium to shift toward the products?

increase P°



shift towards fewer
+ make gas



When 4.00 mol each of $X(g)$ and $Y(g)$ are placed in a 1.00 L vessel and allowed to react at constant temperature according to the equation above, 6.00 mol of $Z(g)$ is produced. What is the value of the equilibrium constant, K_c ?

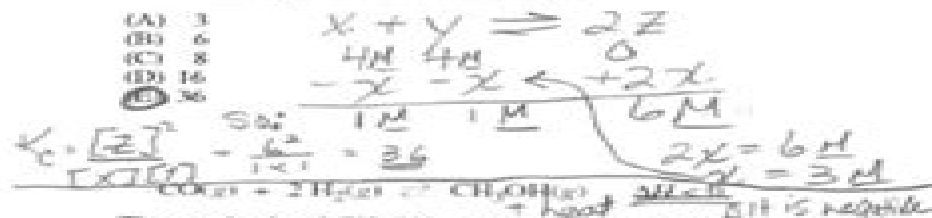
(A) 3

(B) 6

(C) 8

(D) 16

(E) 36



6. The synthesis of $CH_3OH(l)$ from $CO_2(g)$ and $H_2(g)$ is represented by the equation above. The value of K_c for the reaction at 483 K is 14.5.

increase heat shifts left, decrease K_c

Which of the following explains the effect on the equilibrium constant, K_c , when the temperature of the reaction system is increased to 650 K?

(A) K_c will increase because the activation energy of the forward reaction increases more than that of the reverse reaction.(B) K_c will increase because there are more reactant molecules than product molecules.(C) K_c will decrease because the reaction is exothermic.(D) K_c is constant and will not change.

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Princeton Review



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Multiple-Choice and Free-Response Questions in Preparation for the AP Chemistry Examination 5th Edition Peter E. Demmin, 2005 Student's Solutions Manual to Accompany Multiple-choice and Free-Response Questions in Prep for AP Chemistry Examination 5th Edition Peter E. Demmin, David W. Hostage, 2005-01-01 *Oxygen Radicals in Chemistry and Biology* Wolf Bors, Manfred Saran, David Tait, 2011-06-15 No detailed description available for Oxygen Radicals in Chemistry and Biology **Multiple-choice & Free-response Questions in Preparation for the AP Chemistry Examination** Peter E. Demmin, 2005 Pharmaceutical Biotechnology Oliver Kayser, Heribert Warzecha, 2012-03-27 This second edition of a very successful book is thoroughly updated with existing chapters completely rewritten while the content has more than doubled from 16 to 36 chapters As with the first edition the focus is on industrial pharmaceutical research written by a team of industry experts from around the world while quality and safety management drug approval and regulation patenting issues and biotechnology fundamentals are also covered In addition this new edition now not only includes biotech drug development but also the use of biopharmaceuticals in diagnostics and vaccinations With a foreword by Robert Langer Kenneth J. Germeshausen Professor of Chemical and Biomedical Engineering at MIT and member of the National Academy of Engineering and the National Academy of Sciences *Multiple-Choice and Free-Response Questions in Preparation for the AP Chemistry Examination* Peter E. Demmin, 2000-09 Cracking the AP Chemistry Exam, 2014 Edition (Revised) Princeton Review, 2014-03-18 THE PRINCETON REVIEW GETS RESULTS Get all the prep you need to ace the revised AP Chemistry Exam with 2 full length practice tests thorough topic reviews and proven techniques to help you score higher The AP Chemistry course and test are undergoing major changes with a new version of the exam debuting in May 2014 Inside Cracking the AP Chemistry Exam you'll find 2 full length practice tests with detailed explanations that include the new multiple choice and constructed response question types Expert subject reviews for all test topics that reflect the changes to the 2014 AP Chemistry exam including newly incorporated test topics and Big Ideas organization Practice drills at the end of each chapter Step by step strategies techniques for every section of the exam A comprehensive list of key chemistry equations and constants This eBook edition has been specially formatted for on device viewing with cross linked questions answers and explanations *Multiple Choice Questions in Preparation for the AP Chemistry Examination* Peter E. Demmin, 1996-01-01 AP Chemistry 2020 & 2021: AP Chemistry Review Book and Practice Questions for the Advanced Placement Chem Exam Test Prep Books, 2019-11-21 Test Prep Books AP Chemistry 2020 2021 AP Chemistry Review Book and Practice Questions for the Advanced Placement Chem Exam Made by Test Prep Books experts for test takers trying to achieve a great score on the AP Chemistry exam This comprehensive study guide includes Quick Overview Find out what's inside this guide Test Taking Strategies Learn the best tips to help overcome your exam Introduction Get a thorough breakdown of what the test is and what's on it Atomic Structure and Properties Molecular and Ionic Compound Structure and

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