An Activity Series

Purpose: The purpose of this lab is to verify the activity series of five metals and three halogens. This can be done by reacting the metal with various metal solutions to see which of them creates a displacement reaction. As for the halogens, it can be done in a similar way as metal, but with the halogens in their liquid form and ion form. The significance of this experiment is that geologist could use them to extract minerals from ones.

Hypothesis: The hypothesis of this experiment is that the experiment will help verify the activity series of metals and halogens. This is because the reaction would not occur if the substance trying to replace the other substance is less reactive than that other substance. If the reaction did happen, there would be an apparent change in the sample. For example, if one metal was replaced by the other, the solid in the solution would be changed to the metal that was replaced. As for halogens, each halogen possess different color from another and so if one halogen was replaced by the other, the color of the solution would be changed.

Materials:

Quantity
Committee
4 mm ¹
4mL
4 mm ³
4mL
4 mm³
4mL
4mL
4 mm ³
4mL
5 pipettes
24 test tubes
30 labels

Part 2		
Materials	Quantity	
Bromine water (Br ₂)	3mL	
Chlorine water (Cl ₂)	3mL	
lodine water (I ₂)	3mL	
Mineral oil	36mL	
0.1 M Potassium iodide solution (KI)	SmL	
0.1 M Sodium bromide (NaBr)	SmL	
0.1 M Sodium chloride (NaCl)	5mL	
Labeling tape	1 roll	

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Jin-Ying Zhang

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