

CHEMICAL ACTIVITY OF METALS

Lab Guidelines (1,2,3,4,5&6[attach guidelines],7,10,12)

Purposes:

To observe the reactivity of some common metals.

To understand the dynamics of a single replacement reaction, also known as an oxidation/reduction reaction.

To understand why some metals are more useful to us than others.

Introductory Information:

What is a single replacement reaction?

What is an oxidation/reduction reaction? What is going on with the electrons?

What does a more active metal do to a less active one?

What will I be looking for as evidence of chemical activity?

Materials:

Well plate, solutions of $\text{Cu}(\text{NO}_3)_2$, $\text{Mg}(\text{NO}_3)_2$, $\text{Zn}(\text{NO}_3)_2$, AgNO_3

4 pieces of each of the following metals: Cu, Mg, Zn, Ag.

Procedure:

In separate wells, add each metal to several drops of each solution and record your observations. Use the data table below to guide you. If the metals are not shiny, use steel wool to polish them before attempting the reactions.

HINT: One of the reactions is very faint, so make sure you allow 1 minute for a reaction to take place!!

Dispose of contents of well plate in garbage can. Then rinse well plate thoroughly with water.

Data:

Create a data table like the one below ON YOUR OWN PAPER:

Record your observations in this table. Please use the abbreviation "ppt" if a precipitate appears and note the color. If no reaction occurs, write "NR." HINT: One of the reactions is very faint.

	$\text{Cu}(\text{NO}_3)_2$	$\text{Mg}(\text{NO}_3)_2$	$\text{Zn}(\text{NO}_3)_2$	AgNO_3
Cu				
Mg				
Zn				
Ag				

Activity Of Metal Pre Lab Answers

**Rosemary A. Marusak, Kate Doan, Scott
D. Cummings**

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Exploring General Chemistry in the Laboratory Colleen F. Craig, Kim N. Gunnerson, 2017-02-01 This laboratory manual is intended for a two semester general chemistry course The procedures are written with the goal of simplifying a complicated and often challenging subject for students by applying concepts to everyday life This lab manual covers topics such as composition of compounds reactivity stoichiometry limiting reactants gas laws calorimetry periodic trends molecular structure spectroscopy kinetics equilibria thermodynamics electrochemistry intermolecular forces solutions and coordination complexes By the end of this course you should have a solid understanding of the basic concepts of chemistry which will give you confidence as you embark on your career in science

Integrated Approach to Coordination Chemistry Rosemary A. Marusak, Kate Doan, Scott D. Cummings, 2007-03-07 Coordination chemistry is the study of compounds formed between metal ions and other neutral or negatively charged molecules This book offers a series of investigative inorganic laboratories approached through systematic coordination chemistry It not only highlights the key fundamental components of the coordination chemistry field it also exemplifies the historical development of concepts in the field In order to graduate as a chemistry major that fills the requirements of the American Chemical Society a student needs to take a laboratory course in inorganic chemistry Most professors who teach and inorganic chemistry laboratory prefer to emphasize coordination chemistry rather than attempting to cover all aspects of inorganic chemistry because it keeps the students focused on a cohesive part of inorganic chemistry which has applications in medicine the environment molecular biology organic synthesis and inorganic materials

Hands on Physical Science Salvator S. Levy, 1990 **Resources in Education**, 1998

Chemistry Eugene LeMay, Jr., Herbert Beall, Karen M. Robblee, Douglas C. Brower, Prentice-Hall Staff, 2002-02

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Geme III, 2019-04-01 Welcome to the 21st Edition of Nelson Textbook of Pediatrics the reference of choice among pediatricians pediatric residents and others involved in the care of young patients This fully revised edition continues to provide the breadth and depth of knowledge you expect from Nelson while also keeping you up to date with new advances in the science and art of pediatric practice Authoritative and reader friendly it delivers the information you need in a concise easy to use format for everyday reference and study From rapidly changing diagnostic and treatment protocols to new technologies to the wide range of biologic psychologic and social problems faced by children today this comprehensive

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