

Name: \_\_\_\_\_

ANSWER

KEY

Date: \_\_\_\_\_

Period: \_\_\_\_\_

## Activity Series Review Worksheet

### Practice Using the Activity Series

Using the Activity Series Handout, circle the element that is more likely to replace the other in a compound.

Cu Sn or Ba      Ni Al      P or I      Cr or Br         Zn or

### Predicting Single Displacement Reactions

Part 1: Write the chemical name for each compound below.

Part 2: Use the Activity Series to determine if a reaction will occur.

For those reactions that do occur, predict the products and balance the equation.  
If no reaction occurs, write "NR" (No Reaction).

1.  $\text{Mn} + \text{Cr}(\text{NO}_3)_3 \rightarrow$   $\text{Cr} + \text{Mn}(\text{NO}_3)_2$  assume manganese (II)  
Manganese + Chromium (III) Nitrate  $\rightarrow$  Chromium + Manganese (II) Nitrate
2.  $\text{Cu} + \text{ZnSO}_4 \rightarrow$  NR assume copper (II)  
Copper + Zinc Sulfate
3.  $\text{H}_2 + \text{BiCl}_3 \rightarrow$   $2\text{Bi} + 2\text{HCl}$   
Hydrogen + Bismuth Chloride  $\rightarrow$  Bismuth + Hydrogen Chloride
4.  $\text{F}_2 + 2\text{KCl} \rightarrow$   $\text{Cl}_2 + 2\text{KF}$   
Fluorine + Potassium Chloride  $\rightarrow$  Chlorine + Potassium Fluoride
5.  $\text{Fe} + \text{CaSO}_4 \rightarrow$  NR assume iron (III)  
Iron + Calcium sulfate
6.  $\text{Pb} + \text{Zn}(\text{NO}_3)_2 \rightarrow$  NR assume lead (II)  
Lead + Zinc nitrate
7.  $\text{Co} + 2\text{H}_2\text{O} \rightarrow$   $\text{H}_2 + \text{Co}(\text{OH})_2$  assume cobalt (II)  
Cobalt + Dihydrogen Monoxide  $\rightarrow$  Hydrogen + Cobalt (II) Hydroxide
8.  $\text{Zn} + \text{KNO}_3 \rightarrow$  NR  
Zinc + Potassium nitrate
9.  $\text{Ca} + \text{Co}(\text{NO}_3)_2 \rightarrow$   $\text{Co} + \text{Ca}(\text{NO}_3)_2$   
Calcium + Cobaltous Nitrate  $\rightarrow$  Cobalt + Calcium Nitrate
10.  $\text{Sn} + \text{Pb}(\text{NO}_3)_2 \rightarrow$   $\text{Sn}(\text{NO}_3)_2 + \text{Pb}$  assume tin (II)  
Tin + Lead (II) Nitrate  $\rightarrow$  Tin (II) Nitrate + Lead
11.  $\text{Mg} + 2\text{HNO}_3 \rightarrow$   $\text{H}_2 + \text{Mg}(\text{NO}_3)_2$   
Magnesium + Hydrogen Nitrate  $\rightarrow$  Hydrogen + Magnesium Nitrate

# Activity Series Answer Key

**Disha Experts**

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the 'Disha Experts' text.

## Activity Series Answer Key:

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education Moreover the evaluation of physical activity interventions for children is challenging raising a discussion on how to best evaluate these interventions

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## **Table of Contents Activity Series Answer Key**

1. Understanding the eBook Activity Series Answer Key
  - The Rise of Digital Reading Activity Series Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Activity Series Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Activity Series Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Activity Series Answer Key
  - Personalized Recommendations
  - Activity Series Answer Key User Reviews and Ratings
  - Activity Series Answer Key and Bestseller Lists
5. Accessing Activity Series Answer Key Free and Paid eBooks
  - Activity Series Answer Key Public Domain eBooks
  - Activity Series Answer Key eBook Subscription Services
  - Activity Series Answer Key Budget-Friendly Options

6. Navigating Activity Series Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Activity Series Answer Key Compatibility with Devices
  - Activity Series Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Activity Series Answer Key
  - Highlighting and Note-Taking Activity Series Answer Key
  - Interactive Elements Activity Series Answer Key
8. Staying Engaged with Activity Series Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Activity Series Answer Key
9. Balancing eBooks and Physical Books Activity Series Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Activity Series Answer Key
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Activity Series Answer Key
  - Setting Reading Goals Activity Series Answer Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Activity Series Answer Key
  - Fact-Checking eBook Content of Activity Series Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements



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