# **Chemical Bonding Review**

he	em	ical Bonding	Reviev	•	Name:			
rect	ions	: Answer each of the foll	owing ques	tions by circlin	g the correct an	swer choice.		
1	Wil	nich two elements have ti	he same nu	mber of valenc	e electrons?			
	-	Boron and Aluminum	C.	Areon and He	lium			
	8.	Carbon and Nitrogen	D.	Silicon and Al	uminum			
2.	WATE	What type of bond forms when electrons are transferred?						
	A.	Covalent	C.	Metallic				
	8.	Ionic	D.	Nonmetallic				
3.	What is the chemical symbol for a potassium ion?							
		K**		p**1				
	8.	R <sup>-3</sup>	D.	Pr-1				
4.	Ho	w many electrons are ne	eded for a f	illed outermos	t energy level in	the Lewis-dot s	tructure below	
4.	Ho	· Ç ·	eded for a f		t energy level in	the Lewis-dot s	tructure below	
4	A B.	· C·	c.		t energy level in	the Lewis-dot s	tructure below	
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5.	A B. WI A B.	4 0 hen atoms gain electrons Positively charged ions	C. D. during cher C. D.	2 1 mical bonding. Negatively ch Chemically ch	they become arged ions arged ions	the Lewis-dot s	tructure below	
5.	A.B. WI A.B. Po	4 0 hen atoms gain electrons Positively charged ions Neutral ions	C. D. during cher C. D. smple of wh	2 1 mical bonding. Negatively ch Chemically ch	they become arged ions arged ions	the Lewis-dot s	tructure below	
5.	AB WAB POA	4 0 hen atoms gain electrons Positively charged ions Neutral ions tassium chloride is an exa	C. D. during cher C. D. smple of wh	2 1 mical bonding. Negatively ch Chemically ch	they become arged ions arged ions	the Lewis-dot s	tructure below	
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# **Activity 4 Chemical Bonds Answers**

George E. Totten, Rafael Colas

#### **Activity 4 Chemical Bonds Answers:**

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mathematics in our classrooms Daily decisions about how to incorporate creativity choice and autonomy integral components of engagement can build students self efficacy keep them motivated and strengthen their identities as scientists and mathematicians. In this book Eric Brunsell and Michelle A Fleming show you how to apply the joyful learning framework introduced in Engaging Minds in the Classroom to instruction in science and mathematics. Acknowledging that many students particularly girls and students of color do not see themselves as mathematicians and scientists the authors provide a series of suggested activities that are aligned with standards and high expectations to engage and motivate all learners. Given the current focus on encouraging students to pursue science technology engineering and mathematics. STEM studies this book is a welcome addition to every teacher a reference collection Eric Brunsell is a former high school science teacher and is now associate professor of science education at the University of Wisconsin Oshkosh Michelle A Fleming is a former elementary and middle school teacher and is now assistant professor of science and mathematics education at Wright State University in Dayton Ohio

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