

2. textbook-covalent

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Chemistry

Covalent bonding

Learning objectives:

- recognise substances made of small molecules from their formula.
- draw dot and cross diagrams for small molecules
- deduce molecular formula from models and diagrams.

KEY WORDS

covalent
giant covalent
structure
polymer
single bond

When two or more non-metal atoms bond together they share electrons. In this way they form small individual molecules. Such molecules can exist on their own, and don't have to be part of a large structure. They are free to move around so the collection of molecules exists as liquids or gases.

Sharing electrons

When two or more non-metal atoms bond together they form a molecule. Molecules are bonded by covalent bonds. These bonds are very strong. An example is a hydrogen molecule, H_2 . Each atom has a single electron so the two atoms share their electrons as a pair.

This can be represented as:

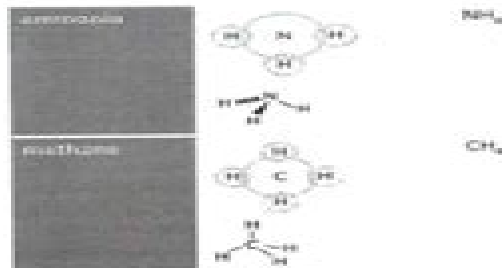
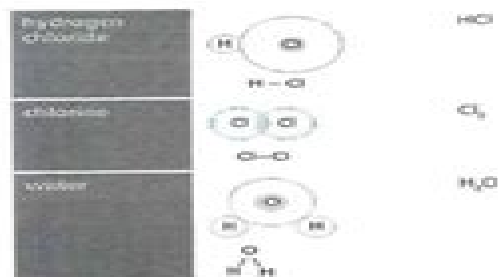


Only one pair of electrons is shared so it is a **single bond**.

- 1 Hydrogen and fluorine bond with a single bond. Draw a diagram to show this.

Covalently bonded molecules

Other small molecules that are covalently bonded are hydrogen chloride, chlorine, water, ammonia and methane. These are bonded by single bonds when a pair of electrons is shared.



DID YOU KNOW?

Some covalently bonded substances may consist of small molecules, but some covalently bonded substances have very large molecules, such as polymers. (Topic 2.9). Some covalently bonded substances have giant covalent structures, such as diamond and silicon dioxide (Topic 2.10).

8 2 The Nature Of Covalent Bonding Textbook

Wolfgang Guggemos



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The book delves into 8 2 The Nature Of Covalent Bonding Textbook. 8 2 The Nature Of Covalent Bonding Textbook is a vital topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into 8 2 The Nature Of Covalent Bonding Textbook, encompassing both the fundamentals and more intricate discussions.

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