

Reversible Reactions & Equilibrium

Question 1

Hydrated cobalt(II) chloride undergoes thermal decomposition when heated to produce anhydrous cobalt(II) chloride and water.



The addition of water to CoCl_2 rehydrates the compound. Which row correctly describes the reverse reaction?

	reaction type	colour change
A	exothermic	blue to pink
B	endothermic	blue to pink
C	exothermic	pink to blue
D	endothermic	pink to blue

[1 mark]

Question 2

Which of the following reactions is reversible?

- A. $\text{Zn} + \text{CuSO}_4 \rightarrow \text{ZnSO}_4 + \text{Cu}$
- B. $\text{CuSO}_4 \cdot 5\text{H}_2\text{O} \rightarrow \text{CuSO}_4 + 5\text{H}_2\text{O}$
- C. $\text{Fe} + \text{H}_2\text{SO}_4 \rightarrow \text{FeSO}_4 + \text{H}_2$
- D. $\text{KCl} + \text{AgNO}_3 \rightarrow \text{KNO}_3 + \text{AgCl}$

[1 mark]

Question 3

Carbon and steam react together as shown in the equation.



Both of the products from the forward reaction are gaseous. Which conditions of pressure and temperature would provide the highest yield of H_2 ?

	pressure	temperature
A	low	low
B	low	high
C	high	low
D	high	high

[1 mark]

18 3 Reversible Reactions And Equilibrium Answer Key

Chryssostomos Chatgililoglu



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Table of Contents 18 3 Reversible Reactions And Equilibrium Answer Key

1. Understanding the eBook 18 3 Reversible Reactions And Equilibrium Answer Key
 - The Rise of Digital Reading 18 3 Reversible Reactions And Equilibrium Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying 18 3 Reversible Reactions And Equilibrium Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 18 3 Reversible Reactions And Equilibrium Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from 18 3 Reversible Reactions And Equilibrium Answer Key
 - Personalized Recommendations
 - 18 3 Reversible Reactions And Equilibrium Answer Key User Reviews and Ratings
 - 18 3 Reversible Reactions And Equilibrium Answer Key and Bestseller Lists
5. Accessing 18 3 Reversible Reactions And Equilibrium Answer Key Free and Paid eBooks
 - 18 3 Reversible Reactions And Equilibrium Answer Key Public Domain eBooks
 - 18 3 Reversible Reactions And Equilibrium Answer Key eBook Subscription Services
 - 18 3 Reversible Reactions And Equilibrium Answer Key Budget-Friendly Options
6. Navigating 18 3 Reversible Reactions And Equilibrium Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - 18 3 Reversible Reactions And Equilibrium Answer Key Compatibility with Devices
 - 18 3 Reversible Reactions And Equilibrium Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 18 3 Reversible Reactions And Equilibrium Answer Key
 - Highlighting and Note-Taking 18 3 Reversible Reactions And Equilibrium Answer Key
 - Interactive Elements 18 3 Reversible Reactions And Equilibrium Answer Key
8. Staying Engaged with 18 3 Reversible Reactions And Equilibrium Answer Key

- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 18 3 Reversible Reactions And Equilibrium Answer Key
9. Balancing eBooks and Physical Books 18 3 Reversible Reactions And Equilibrium Answer Key
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection 18 3 Reversible Reactions And Equilibrium Answer Key
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine 18 3 Reversible Reactions And Equilibrium Answer Key
- Setting Reading Goals 18 3 Reversible Reactions And Equilibrium Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 18 3 Reversible Reactions And Equilibrium Answer Key
- Fact-Checking eBook Content of 18 3 Reversible Reactions And Equilibrium Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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