

Name: _____

Student ID: _____

Circle your Discussion Time: R 9-10 F 8-9 F 1-2

***"Just for fun" means just for fun on the quiz, but will help you study for the test.
(red answers were graded, blue ones were just for fun)

There was only one version to the quiz! Tee hee hee.

1. What's the shape name and approximate bond angle for H_2O ? (3 points)

Shape name

bent

Approximate bond angle

< 109.5°

Just for fun, tell me at what temperature is water most dense.

4°C

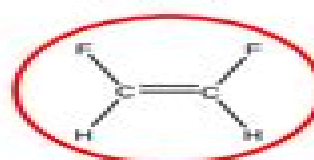
2. What's the shape (name) of ammonia, NH_3 ? (2 points)

trigonal pyramid

Just for fun, what's the geometry and approximate bond angles?

tetrahedral, ~109.5°

3. Circle the molecules that have a dipole moment: (1 point)



Just for fun, rank the polar molecules in order of increasing polarity.



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>



>



~

4. For carbon dioxide, list all the bonding orbitals and indicate in which bonds they are involved **for the carbon atom**. (3 points)



bonds they are

2 (2sp), 2 (2p)

Just for fun, what's the hybridization and bond angles for the remaining atoms?

3 (2sp²), 2p for both oxygen atoms

5. For the following molecule, state the hybridization at the specified oxygen and carbon atoms: (3 points)

14a Atomic Structure Answers

Disha Experts

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'Disha Experts' text.

14a Atomic Structure Answers:

(Chemistry) Inorganic Chemistry: Atomic Structure, Chemical Bonding and Fundamentals of Organic

Chemistry Dr. Mohd. Irfan Ahmad Khan, 2020-03-19 Buy Latest Chemistry Inorganic Chemistry Atomic Structure Chemical Bonding and Fundamentals of Organic Chemistry in English language for B Sc 1st Semester Bihar State By Thakur publication

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Review of Medical Dosimetry William Amestoy, 2015-03-03 This study guide will be a reliable support and easy to use source of information for students in the fields of dosimetry physics radiation oncology and therapy as they progress through the educational levels in preparation for board examinations The theoretical and practical knowledge gained by students on previous courses or in clinical settings is reinforced by means of almost 1200 questions and accompanying detailed analytical answers In order to cater for the needs of all students the questions are arranged according to three levels of difficulty The level 1 questions are mainly intended for those hoping to pass the Medical Dosimetrist Certification Board MDCB exam but will also be beneficial for Medical Physics candidates taking written exams and for Radiation Oncology residents The level II questions are in general clinically related and will be relevant for any

student while the level III questions are advanced and are especially suitable for American Board of Radiology candidates or those taking equivalent exams elsewhere in the world The study guide is broken down into different subject areas with provision of multiple questions and answers on each subject In addition the mathematical and physics questions include brief explanations of how the student can solve each problem At the end of the guide three practice tests are included with the same number of questions as are found in the MDCB exam These tests will help students to test their knowledge and improve their test taking speed

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Kota,2021-01-13 Chemistry for NEET Volume 1 Class XI is designed to serve the requirements of medical aspirants preparing for NEET in the best possible manner Through the course of this book the aspirants have been provided with a pedagogically set problems to help them prepare for these examinations better Instead of chasing their mentors for concept based questions on a regular basis the aspirants can now practice whenever they wish to and absolutely on their own Questions in this book are handpicked by experienced faculty members of Career Point to enhance the following skills of the students 1 Understanding of concepts and their application to the grass root level 2 Improving their scoring ability accuracy by providing an opportunity to practice a variety of questions Features of Book are 2200 Questions with explanatory Solutions Chapters according to NCERT All Types of MCQs based on latest pattern Previous Year Questions since 2005 3 Mock Tests for Final Touch

Foundations of Reasoning under Uncertainty Bernadette Bouchon-Meunier,Luis Magdalena,Manuel

Ojeda-Aciego,José-Luis Verdegay,Ronald R. Yager,2010-01-11 Uncertainty exists almost everywhere except in the most idealized situations it is not only an inevitable and ubiquitous phenomenon but also a fundamental scientific principle Furthermore uncertainty is an attribute of information and usually decision relevant information is uncertain and or imprecise therefore the abilities to handle uncertain information and to reason from incomplete knowledge are crucial features of intelligent behaviour in complex and dynamic environments By carefully exploiting our tolerance for imprecision and approximation we can often achieve tractability robustness and better descriptions of reality than traditional deductive methods would allow us to obtain In conclusion as we move further into the age of machine intelligence

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