

2009 AP[®] CHEMISTRY FREE-RESPONSE QUESTIONS



3. Methane gas reacts with chlorine gas to form dichloromethane and hydrogen chloride, as represented by the equation above.
- (a) A 25.0 g sample of methane gas is placed in a reaction vessel containing 2.58 mol of $\text{Cl}_2(g)$.
- Identify the limiting reactant when the methane and chlorine gases are combined. Justify your answer with a calculation.
 - Calculate the total number of moles of $\text{CH}_2\text{Cl}_2(g)$ in the container after the limiting reactant has been totally consumed.

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Mary Millhollon, Richard H. Langley



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