

## PERIODIC TABLE TRENDS

**Step 1:** Complete the squares for each element by adding the atomic number, name, and atomic mass.

Write the atomic number at the top of the square.

Write the atomic name under the symbol.

Write the atomic mass at the bottom of the square.

**Step 2:** Determine the number of electrons, protons, and neutrons in each statement.

**Step 3:** Create a Bohr diagram for each element.

**Step 4:** Draw a Lewis Dot Structure for each element.

**Step 5:** Show whether the element is a metal, nonmetal, or metalloid by circling the correct response. Also determine if the element is a solid or gas by circling the correct choice.

**Step 6:** Use the following colors to shade in the square for each element. You should **ONLY** color in the small square in the upper left-hand corner and not the entire card.

Green – Li and Na

Pink – O and S

Blue – Be and Mg

Purple – F and Cl

Orange – B and Al

Red – C and Si

Brown – N and P

Yellow – He, Ne, and Ar

White – H

**Step 7:** Cut the cards apart and arrange according to atomic number in the pattern shown below. Once you have the cards arranged in the correct order, glue them to a large sheet of construction paper.

1							2
3	4	5	6	7	8	9	10
11	12	13	14	15	16	17	18

**Step 8:** Put a title on your table.

**Step 9:** Make a key for each color with the name of the group for that column.

**Step 10:** Write the column number at the top of each group and the period number on the left side of the table. Be careful, column numbers are not in order from 1-18 and the transition metals, columns 3-12 are not on the table.

**Step 11:** Draw the red zigzag line that separates the metals from the nonmetals.

**Step 12:** Answer the questions on the worksheet using the information on your periodic table.

Be

P = \_\_\_\_  
 N = \_\_\_\_  
 E = \_\_\_\_



Bohr Diagram



Dot Structure

Metal

Solid

Nonmetal

Gas

Metalloid

Gas

## 2 Exploring The Periodic Table Answer Key

**Roger Norris**



## 2 Exploring The Periodic Table Answer Key:

*Exploring Physical Science in the Laboratory* John T. Salinas ,2019-02-01 This full color manual is designed to satisfy the content needs of either a one or two semester introduction to physical science course populated by nonmajors It provides students with the opportunity to explore and make sense of the world around them to develop their skills and knowledge and to learn to think like scientists The material is written in an accessible way providing clearly written procedures a wide variety of exercises from which instructors can choose and real world examples that keep the content engaging Exploring Physical Science in the Laboratory guides students through the mysteries of the observable world and helps them develop a clear understanding of challenging concepts *FCS physical science L2 ,2007* **Class 12 Chemistry CBSE Board 100**

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