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THE BEHAVIOR OF GASES

Chapter Quiz

Indicate whether the statement is true or false.

1. Boyle's law states that, at constant temperature, the volume of a fixed amount of gas is inversely proportional to the pressure.
2. According to Charles's law, the volume of a fixed amount of gas is directly proportional to the absolute temperature.
3. Gay-Lussac's law states that, at constant volume, the pressure of a fixed amount of gas is directly proportional to the absolute temperature.
4. The combined gas law states that, for a fixed amount of gas, the product of the pressure and volume is proportional to the absolute temperature.
5. The ideal gas law states that, for a fixed amount of gas, the product of the pressure and volume is proportional to the absolute temperature.
6. The ideal gas law states that, for a fixed amount of gas, the product of the pressure and volume is proportional to the absolute temperature.

Indicate whether the statement is true or false.

7. The ideal gas law states that, for a fixed amount of gas, the product of the pressure and volume is proportional to the absolute temperature.

Answer Key

1. True
2. True
3. True
4. True
5. True
6. True

Answer Key

7. True

1. Boyle's law states that, at constant temperature, the volume of a fixed amount of gas is inversely proportional to the pressure.	True
2. According to Charles's law, the volume of a fixed amount of gas is directly proportional to the absolute temperature.	True
3. Gay-Lussac's law states that, at constant volume, the pressure of a fixed amount of gas is directly proportional to the absolute temperature.	True
4. The combined gas law states that, for a fixed amount of gas, the product of the pressure and volume is proportional to the absolute temperature.	True
5. The ideal gas law states that, for a fixed amount of gas, the product of the pressure and volume is proportional to the absolute temperature.	True
6. The ideal gas law states that, for a fixed amount of gas, the product of the pressure and volume is proportional to the absolute temperature.	True

True

True

True

1the Behavior Of Gases Chapter Quiz

Norman C. Harris



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Instructor's Manual to Accompany Introductory Applied Physics, Third Edition Norman C. Harris, 1972

Fundamentals of Ionized Gases Boris M. Smirnov, 2012-09-19 A comprehensive and readily accessible work for studying the physics of ionized gases based on *Physics of Ionized Gases* The focus remains on fundamentals rather than on the details required for interesting but difficult applications such as magnetic confinement fusion or the phenomena that occur with extremely high intensity short pulse lasers However this new work benefits from much rearranging of the subject matter within each topic resulting in a more coherent structure There are also some significant additions many of which relate to clusters while other enlarged sections include plasmas in the atmosphere and their applications In each case the emphasis is on a clear and unified understanding of the basic physics that underlies all plasma phenomena Thus there are chapters on plasma behavior from the viewpoint of atomic and molecular physics as well as on the macroscopic phenomena involved in physical kinetics of plasmas and the transport of radiation and of charged particles within plasmas With this grounding in the fundamental physics of plasmas the notoriously difficult subjects of nonlinear phenomena and of instabilities in plasmas can then be treated with comprehensive clarity The work is rounded off with appendices containing information and data of great importance and relevance that are not easily found in other books Valuable reading for graduate and PhD

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