

# Introduction to Chemistry

## Section 1.1 The Stories of Two Chemicals

In your notebook, read about the ozone layer.

Use each of the boxes below just once to complete the passage.

|              |              |                       |            |
|--------------|--------------|-----------------------|------------|
| atmosphere   | integrated   | ozone                 | ozone hole |
| stratosphere | stratosphere | ultraviolet radiation |            |

Earth's (90) \_\_\_\_\_ is made up of several layers. The air we breathe makes up the lowest layer. This layer is called the (80) \_\_\_\_\_. The next layer up is called the (82) \_\_\_\_\_. This layer contains a protective (84) \_\_\_\_\_ layer.

Closest to us is the (86) \_\_\_\_\_, is made by ultraviolet radiation in the upper part of the stratosphere. The ozone hole is a layer around Earth, with a (88) \_\_\_\_\_. Without ozone, you are more likely to get a melanoma or possibly skin cancer. The covering of the ozone layer is called the (92) \_\_\_\_\_. In developed nations, without ozone will experience an Earth air called an ozone hole or ozone depletion.

In your notebook, read about chlorofluorocarbons.

Use each statement below, write in or follow.

- \_\_\_\_\_ (9) CFC is another name for a chlorofluorocarbon.
- \_\_\_\_\_ (10) CFCs are made up of carbon, fluorine, and chlorine.
- \_\_\_\_\_ (11) All CFCs are synthetic chemicals.
- \_\_\_\_\_ (12) CFCs usually react easily with other chemicals.
- \_\_\_\_\_ (13) CFCs were designed as replacements for toxic refrigerants.

# 1stoichiometry Study Guide For Mastery Content Answers

**Ying-Ying Zheng**



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