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HONORS CHEM 1st SEMESTER EXAM STUDY GUIDE FALL 2010

STRATEGY: Start by reading through your notes to refresh your memory on these topics. Then, use this review sheet as a **starting point** to identify the areas on which you need to spend more study time. For those areas, go back to homework assignments, quizzes, and reviews to practice more problems. I would also recommend going through all of your tests since these questions are only samples and do not include specific examples of how vocabulary and other conceptual information might appear in a multiple-choice or other format. Remember you can access notes and reviews on my website.

IMPORTANT INFORMATION:

- This must be handed in BEFORE taking your final exam
- If **entirely** complete, 5 points will be added to your exam grade
- **All** work must be shown in **DETAIL** on lined paper

Introduction to Chemistry – Chapter 1

1. Identify safe/unsafe lab practices.
2. List pieces and uses of lab equipment.
3. List the requirements for tables and graphs.
4. What is plotted on the x-axis and on the y-axis?
5. Make a density graph.
6. What is water displacement?

Matter and Change – Chapter 2

7. Classify the following substances as *solid*, *liquid*, *gas*, *or plasma* based on their properties.
   a. flexible volume, high KE, particles can disperse freely.
   b. flexible volume, very high KE, particles are charged.
   c. fixed volume, very low KE, orderly particles.
   d. fixed volume, low KE, particles can move past each other.
8. Classify the following as element, compound, heterogeneous mixture, or solution.
   a. graphite (carbon)
   b. grape juice
   c. table salt (NaCl)
   d. pepper
9. Classify the following as *chemical* or *physical changes*.
   a. cutting wire
   b. ripening tomato
   c. apple slices turning brown
   d. compressing a gas
10. Classify the following properties as *physical* or *chemical*.
    a. melts at 68.0°C
    b. corrosive
    c. reacts violently with water
    d. decomposes in air
    e. magnetic
11. Classify the following properties as *intensive* or *extensive*.
    a. melting point
    b. volume
    c. mass
    d. reactivity
    e. density

VOCAB:kinetic molecular theory

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Organic Chemistry Study Guide with Solutions Manual Neil E. Schore,2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer Chemical Energetics, Equilibria and Functional Group Organic Chemistry I - Laboratory Mr. Rohit Manglik,2024-03-03 EduGorilla Publication is a trusted name in the education sector committed to empowering learners with high quality study materials and resources Specializing in competitive exams and academic support EduGorilla provides comprehensive and well structured content tailored to meet the needs of students across various streams and levels Resources in Education ,1998-10 Organic Chemistry, 12e Binder Ready Version Study Guide & Student Solutions Manual T. W. Graham Solomons,Craig B. Fryhle,Scott A. Snyder,2016-04-11 This is the Student Study Guide Solutions Manual to accompany Organic Chemistry 12th Edition The 12th edition of Organic Chemistry continues Solomons Fryhle Snyder s tradition of excellence in teaching and preparing students for success in the organic classroom and beyond A central theme of the authors approach to organic chemistry is to emphasize the relationship between structure and reactivity To accomplish this the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms The authors philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters The structural aspects of the authors approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us

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