

2.1

CHEMICAL REACTIONS

Section Review

Objectives

1. Measure enthalpy change using a calorimeter/bomb
2. Identify and predict chemical reactions using phase rule, plus 2
3. Apply the law of conservation of energy to chemical reactions

Vocabulary

- | | | |
|-----------------------|---------------|----------------------------|
| 1. endothermic | 2. exothermic | 3. enthalpy |
| 4. formation reaction | 5. product | 6. thermochemical reaction |

Part A Completion

Use the information available to identify a thermochemical reaction, balance, and write energy change equation, correct the balance, write equation within time, identify limits for reaction

- | | |
|--|----------------------|
| 1. A chemical change always involves either a change in <u>1</u> | 1. <u>enthalpy</u> |
| 2. A chemical change involves a change in the number of <u>2</u> | 2. <u>atoms</u> |
| 3. A chemical change involves a change in the number of <u>3</u> | 3. <u>molecules</u> |
| 4. A chemical change involves a change in the number of <u>4</u> | 4. <u>moles</u> |
| 5. A chemical change involves a change in the number of <u>5</u> | 5. <u>atoms</u> |
| 6. A chemical change involves a change in the number of <u>6</u> | 6. <u>molecules</u> |
| 7. A chemical change involves a change in the number of <u>7</u> | 7. <u>moles</u> |
| 8. A chemical change involves a change in the number of <u>8</u> | 8. <u>atoms</u> |
| 9. A chemical change involves a change in the number of <u>9</u> | 9. <u>molecules</u> |
| 10. A chemical change involves a change in the number of <u>10</u> | 10. <u>moles</u> |
| 11. A chemical change involves a change in the number of <u>11</u> | 11. <u>atoms</u> |
| 12. A chemical change involves a change in the number of <u>12</u> | 12. <u>molecules</u> |
| 13. A chemical change involves a change in the number of <u>13</u> | 13. <u>moles</u> |
| 14. A chemical change involves a change in the number of <u>14</u> | 14. <u>atoms</u> |
| 15. A chemical change involves a change in the number of <u>15</u> | 15. <u>molecules</u> |
| 16. A chemical change involves a change in the number of <u>16</u> | 16. <u>moles</u> |
| 17. A chemical change involves a change in the number of <u>17</u> | 17. <u>atoms</u> |
| 18. A chemical change involves a change in the number of <u>18</u> | 18. <u>molecules</u> |
| 19. A chemical change involves a change in the number of <u>19</u> | 19. <u>moles</u> |
| 20. A chemical change involves a change in the number of <u>20</u> | 20. <u>atoms</u> |

Part B True-False

Identify each of the following as true or false, or as neither true nor false, or as both true and false

1. A chemical change always involves a change in 1
2. A chemical change always involves a change in 2
3. A chemical change always involves a change in 3
4. A chemical change always involves a change in 4
5. A chemical change always involves a change in 5
6. A chemical change always involves a change in 6
7. A chemical change always involves a change in 7
8. A chemical change always involves a change in 8
9. A chemical change always involves a change in 9
10. A chemical change always involves a change in 10
11. A chemical change always involves a change in 11
12. A chemical change always involves a change in 12
13. A chemical change always involves a change in 13
14. A chemical change always involves a change in 14
15. A chemical change always involves a change in 15
16. A chemical change always involves a change in 16
17. A chemical change always involves a change in 17
18. A chemical change always involves a change in 18
19. A chemical change always involves a change in 19
20. A chemical change always involves a change in 20

1describing Chemical Reactions Section Review

Michael J. Padilla



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