

# 2.1

## CHEMICAL REACTIONS

### Section Review

#### Objectives

1. Measure enthalpy using a calorimeter.
2. Identify and predict chemical reactions using phase diagrams.
3. Apply the law of conservation of energy to chemical reactions.

#### Vocabulary

- |                 |               |                                  |
|-----------------|---------------|----------------------------------|
| 1. endothermic  | 2. exothermic | 3. enthalpy                      |
| 4. fusion curve | 5. product    | 6. law of conservation of energy |

### Part A Completion

Use the vocabulary words to identify a vocabulary word that matches each given description. Write the word in the space provided.

- |   |           |
|---|-----------|
| 1. A process that absorbs energy is called a(n) _____.  | 1. _____  |
| 2. A process that releases energy is called a(n) _____. | 2. _____  |
| 3. The amount of heat in a system is called _____.      | 3. _____  |
| 4. The amount of heat in a system is called _____.      | 4. _____  |
| 5. The amount of heat in a system is called _____.      | 5. _____  |
| 6. The amount of heat in a system is called _____.      | 6. _____  |
| 7. The amount of heat in a system is called _____.      | 7. _____  |
| 8. The amount of heat in a system is called _____.      | 8. _____  |
| 9. The amount of heat in a system is called _____.      | 9. _____  |
| 10. The amount of heat in a system is called _____.     | 10. _____ |
| 11. The amount of heat in a system is called _____.     | 11. _____ |
| 12. The amount of heat in a system is called _____.     | 12. _____ |
| 13. The amount of heat in a system is called _____.     | 13. _____ |
| 14. The amount of heat in a system is called _____.     | 14. _____ |
| 15. The amount of heat in a system is called _____.     | 15. _____ |
| 16. The amount of heat in a system is called _____.     | 16. _____ |
| 17. The amount of heat in a system is called _____.     | 17. _____ |
| 18. The amount of heat in a system is called _____.     | 18. _____ |
| 19. The amount of heat in a system is called _____.     | 19. _____ |
| 20. The amount of heat in a system is called _____.     | 20. _____ |

### Part B True-False

Write true or false in the space provided. If false, write the correct word or phrase in the space provided.

1. The amount of heat in a system is called enthalpy. ☐
2. The amount of heat in a system is called enthalpy. ☐
3. The amount of heat in a system is called enthalpy. ☐

# 1describing Chemical Reactions Section Review Worskeet Answers

**Rex M. Heyworth**



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