

**Lesson Outline****LESSON 2*****Types of Chemical Reactions*****A. Patterns in Reactions**

1. There are \_\_\_\_\_ major types of chemical reactions.
2. Each type of chemical reaction follows a unique \_\_\_\_\_ in the way atoms in reactants \_\_\_\_\_ to form products.

**B. Types of Chemical Reactions**

1. Knowing the types of chemical reactions helps \_\_\_\_\_ how substances will react and what \_\_\_\_\_ will form.
2. In a(n) \_\_\_\_\_ reaction, two or more substances combine.
  - a. The product in a synthesis reaction is a(n) \_\_\_\_\_.
  - b. A synthesis reaction has \_\_\_\_\_ reactant(s) and \_\_\_\_\_ product(s).
3. In a(n) \_\_\_\_\_ reaction, a substance breaks down and forms two or more substances.
  - a. The reactant in a decomposition reaction is a(n) \_\_\_\_\_.
  - b. The products in a decomposition reaction can be \_\_\_\_\_ or compounds.
  - c. A decomposition reaction has \_\_\_\_\_ reactant(s) and \_\_\_\_\_ product(s).
4. In a(n) \_\_\_\_\_ reaction, an atom or group of atoms replaces part of a(n) \_\_\_\_\_.
5. One type of replacement reaction is \_\_\_\_\_ replacement.
  - a. In a(n) \_\_\_\_\_ reaction, one \_\_\_\_\_ replaces another element in a compound.
  - b. The reactants in this type of reaction are a(n) \_\_\_\_\_ and a(n) \_\_\_\_\_.
  - c. The products in this type of reaction are a different \_\_\_\_\_ and a different \_\_\_\_\_.
6. The other type of replacement reaction is \_\_\_\_\_ replacement.
  - a. In a(n) \_\_\_\_\_ reaction, the \_\_\_\_\_ ions in two compounds switch places, forming two new compounds.

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