

Acids & Bases Review

1. Name the following compounds as acids and circle which are weak acids:



nitric acid



sulfuric acid



hydrofluoric acid



carbonic acid



acetic acid



phosphoric acid

2. Write the formulas for the following bases:

a) calcium hydroxide



b) aluminum hydroxide



c) ammonium hydroxide



d) lithium hydroxide



3. Complete the following table:

	Acids	Bases
Definition	A substance that increases the concentration of H^+ (or H_3O^+) when added to water. or A substance that acts as a proton (H^+) donor.	A substance that increases the concentration of OH^- when added to water. or A substance that acts as a proton (H^+) acceptor.
Electrolyte in aqueous solution?	H^+	OH^-
pH	less than 7	greater than 7
Taste	sour	bitter
Litmus turns	blue turns red	red turns blue
Which reacts with metals to form H_2 gas?	yes	no

Conjugate acid-base pairs:

4. Identify the acid, the base, the conjugate acid and the conjugate base in the reactions:



5. An acid will form the conjugate base and a base will form the conjugate acid.

Answers Acids And Bases

Kristi Lew



Answers Acids And Bases:

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Structure in Protein Chemistry Jack Kyte, 2006-11-01 The second edition of Structure in Protein Chemistry showcases the latest developments and innovations in the field of protein structure analysis and prediction The

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I. A Study of the Acid-base Equilibria of Arsphenamine Solutions Elias Elvove, 1924

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web although it was the second to last play that august wilson wrote gem of the ocean is chronologically speaking the first play in his pittsburgh cycle which included ten plays that explored the black american experience in the 20th century

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web gem of the ocean she had her hair slicked back her mouth made her face look pretty she was dancing and she had tears

in her eyes i asked her why she was crying she said she was lonesome i told her i couldn t fix that but if she wanted somebody to walk her home after the dance i d walk her see that she got home safe

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