1989 Advanced Placement Exam

Part I Multiple Choice

Note: For all questions referring to solutions, assume that the solvent is water unless otherwise stated.

<u>Directions:</u> Each set of lettered choices below refers to the numbered statements immediately following it. Select the one lettered choice that best fits each statement and then blacken the corresponding space on the answer sheet. A choice may be used once, more than once, or not at all in each set.

Questions 1-3

a. O	c. Rb	e. N
b. La	d. Mg	

- What is the most electronegative element?
- Which element exhibits the greatest number of different oxidation states?
- 3. Which of the elements above has the smallest ionic radius for its most commonly found ion?

Ouestions 4-7

- An impossible electronic configuration
- The ground-state configuration for the atoms of a transition element
- The ground–state configuration of a negative ion of a halogen
- The ground-state configuration of a common ion of an alkaline earth element

Questions 8-10

- a. a solution with a pH less than 7 that is not a buffer solution
- a buffer solution with a pH between 4 and 7
- c. a buffer solution with a pH between 7 and 10
- a solution with a pH greater than 7 that is not a buffer solution
- e. a solution with a pH of 7

(Ionization constants:

$$CH_3COOH = 1.8 \times 10^{-5};$$

 $NH_3 = 1.8 \times 10^{-5};$
 H_2CO_3 : $K_1 = 4 \times 10^{-7};$
 $K_2 = 4 \times 10^{-11};$

- A solution prepared to be initially 1 M in NaCl and 1 M in HCl.
- A solution prepared to be initially 1 M in Na₂CO₃ and 1 M in CH₃COONa
- A solution prepared to be initially 0.5 M in CH₃COOH and 1 M in CH₃COONa

Questions 11-14

- a. hydrogen bonding
- b. hybridization
- c. ionic bonding
- d. resonance
- e. van der Waals forces (London dispersion forces)
- Is used to explain why iodine molecules are held together in the solid state
- Is used to explain why the boiling point of HF is greater than the boiling point of HBr
- Is used to explain the fact that the four bonds in methane are equivalent
- Is used to explain the fact that the carbon-to-carbon bonds in benzene, C₆H₆, are identical

ĸ.

<u>Directions</u>: Each of the questions or incomplete statements below is followed by five suggested answers or compilations. Select the one that is best in each case and then blacken the corresponding space on the answer sheet.

1989 Ap Chemistry Multiple Choice

Rosina Ehmann

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Understanding Psychological Processes that Underlie Test Takers' Choices of Constructed Response Items Xiang Bo Guiding Conceptions and Assessment Principles for the Praxis Series Carol Anne Dwyer, Lawrence J. Wang, 1999 Stricker, Nicholas T. Longford, Paul W. Holland, Rebecca Zwick, Xiang-Bo Wang, 1993 The Relationship Between Item Format and Cognitive Processes in Wide-scale Assessment of Mathematics Diane R. Garavaglia, 2001 The Cumulative Book Index ,1990 A world list of books in the English language **Uneducated Guesses** Howard Wainer, 2011-08-28 What statistical evidence shows us about our misguided educational policies Uneducated Guesses challenges everything our policymakers thought they knew about education and education reform from how to close the achievement gap in public schools to admission standards for top universities In this explosive book Howard Wainer uses statistical evidence to show why some of the most widely held beliefs in education today and the policies that have resulted are wrong He shows why colleges that make the SAT optional for applicants end up with underperforming students and inflated national rankings and why the push to substitute achievement tests for aptitude tests makes no sense Wainer challenges the thinking behind the enormous rise of advanced placement courses in high schools and demonstrates why assessing teachers based on how well their students perform on tests a central pillar of recent education reforms is woefully misguided He explains why college rankings are often lacking in hard evidence why essay questions on tests disadvantage women why the most grievous errors in education testing are not made by testing organizations and much more No one concerned about seeing our children achieve their full potential can afford to ignore this book With forceful storytelling wry insight and a wealth of real world examples Uneducated Guesses exposes today s educational policies to the light of empirical evidence and offers solutions for fairer and more viable future policies On Giving Test Takers a Choice Among Constructive Response Times Xiang Bo Wang, 1999 **Books in Print** .1991 How to Prepare for the AP Chemistry Neil D. Jespersen, Neil Jespersen, PH. D.,1999-02 This manual offers two diagnostic exams to help students pinpoint their strengths and weaknesses plus three full length practice exams closely modeled on actual AP chemistry exams Energy Research Abstracts ,1990

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