- 6. What type of bond(s) were formed in the compound from question #4? Ionic Bond
- 7. Draw the bond(s) that will form between P and Cl.

:cecopooci: PCl3

8. What is the formula for the compound formed in question #7?

PCL3

9. What type of bond(s) were formed in the compound from question #7?

Covalent Bond

10. Draw the correct valence electron configuration for each atom given.

Ĥ	- <u>T</u> :	Ča -	Al -
Ne:	- Ge	B -	ĸ
ű	si -	- ċ -	Xe :

11. Fill in the table with the correct values.

Element	# valence electrons	# covalent bonds able to be formed
H	2	1
He	a	0
С	4	4
CI	7	7
Ar	8	0
0	6	2
N	5	3
P	5	3

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Eric Lichtfouse, Subramanian Senthilkannan Muthu, Ali Khadir

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