

Name \_\_\_\_\_

**SHORT ANSWER.** Write the word or phrase that best completes each statement or answers the question.

- 1) What is the solvent in an aqueous solution?

**ESSAY.** Write your answer in the space provided or on a separate sheet of paper.

- 2) Explain why water is a good solvent for ionic substances.

**MULTIPLE CHOICE.** Choose the one alternative that best completes the statement or answers the question.

- 3) Of the species below, only \_\_\_\_\_ is
- not
- an electrolyte.

A) HCl                      B) Ar                      C) NaCl                      D) KOH                      E)  $\text{Rb}_2\text{SO}_4$ 

- 4) What is the molarity of an aqueous solution containing 75.3 g of glucose (
- $\text{C}_6\text{H}_{12}\text{O}_6$
- ) in 35.5 mL of solution?

A) 0.197                      B) 2.12                      C) 3.52                      D) 1.85                      E) 11.8

- 5) The concentration of species in 500 mL of a 2.104 M solution of sodium sulfate is \_\_\_\_\_ M sodium ion and \_\_\_\_\_ M sulfate ion.

A) 4.208, 2.104                      B) 1.052, 1.052                      C) 2.104, 1.052                      D) 2.104, 4.208                      E) 2.104, 2.104

- 6) What is the molarity of a solution prepared by diluting 43.72 mL of 1.005 M aqueous
- $\text{K}_2\text{Cr}_2\text{O}_7$
- to 500 mL?

A) 0.879                      B) 87.9                      C) 0.0879                      D) 0.0115                      E) 0.0218

- 7) When aqueous solutions of \_\_\_\_\_ are mixed, a precipitate forms.

A) NaI and KBr  
B)  $\text{Li}_2\text{CO}_3$  and CsI  
C)  $\text{K}_2\text{SO}_4$  and  $\text{CrCl}_3$   
D)  $\text{NiBr}_2$  and  $\text{AgNO}_3$   
E) KOH and  $\text{Ba}(\text{NO}_3)_2$ 

- 8) What are the spectator ions in the reaction between KOH (aq) and
- $\text{HNO}_3$
- (aq)?

A)  $\text{H}^+$  and  $\text{NO}_3^-$   
B)  $\text{OH}^-$  only  
C)  $\text{K}^+$  and  $\text{NO}_3^-$   
D)  $\text{K}^+$  and  $\text{H}^+$   
E)  $\text{H}^+$  and  $\text{OH}^-$ **SHORT ANSWER.** Write the word or phrase that best completes each statement or answers the question.

- 9) What is the concentration of HCl in a 0.100 M solution of HCl? Why?

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