

**AP BIOLOGY CHAPTER 2 WORKSHEET****MULTIPLE CHOICE.** 25 pts.

*Place the letter of the choice that best completes the statement or answers the question in the blank.*

- \_\_\_\_\_ 1. About 25 of the 92 natural elements are known to be essential to life. Which four of these 25 elements make up approximately 96% of living matter?
- carbon, sodium, chlorine, nitrogen
  - carbon, sulfur, phosphorus, hydrogen
  - oxygen, hydrogen, calcium, sodium
  - carbon, hydrogen, nitrogen, oxygen
  - carbon, oxygen, sulfur, calcium
- \_\_\_\_\_ 2. Trace elements are those required by an organism in only minute quantities. Which of the following is a trace element that is required by humans and other vertebrates?
- nitrogen
  - calcium
  - iodine
  - sodium
  - phosphorus
- \_\_\_\_\_ 3. Which of the following statements is *false*?
- Atoms of the various elements differ in their number of subatomic particles.
  - All atoms of a particular element have the same number of protons in their nuclei.
  - The neutrons and protons present in the nucleus of an atom are almost identical in mass; each has a mass of about 1 dalton.
  - An atom is the smallest unit of an element that still retains the properties of the element.
  - Protons and electrons are electrically charged particles. Protons have one unit of negative charge, and electrons have one unit of positive charge.
- \_\_\_\_\_ 4. Each element is unique and different from other elements because of the number of protons in the nuclei of its atoms. Which of the following indicates the number of protons in an atom's nucleus?
- atomic mass
  - atomic weight
  - atomic number
  - mass weight
  - mass number
- \_\_\_\_\_ 5. Calcium has an atomic number of 20 and an atomic mass of 40. Therefore, a calcium atom must have
- 20 protons.
  - 40 electrons.
  - 40 neutrons.
  - A and B only
  - A, B, and C

# Ap Biology Chapter 2answers

**The Princeton Review**



## **Ap Biology Chapter 2answers:**

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