
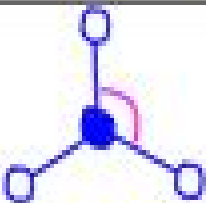
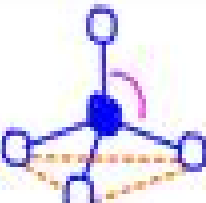
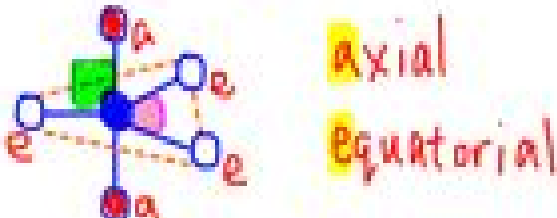
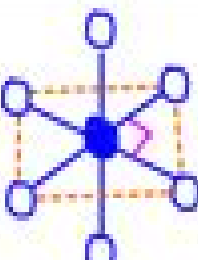


Total no. of electron pairs = bond pairs + lone pairs	Basic Shape	Diagram	Bond angle
2	linear		180°
3	trigonal planar		120°
4	tetrahedral		109.5°
5	trigonal bipyramidal	 <p>axial equatorial</p>	120° 90°
6	octahedral		90°

Answers To Vsepr Model Molecular Shape

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