

## Test Review

Name: \_\_\_\_\_

## AP chemistry Chapter 3

Let's see about reviewing the concepts we've learned in this unit!

1. Consider this mass spectrum for the element boron to the right.



a. How many naturally occurring isotopes does boron have?

b. Calculate the atomic mass of boron— check your answer with the atomic mass on the periodic table.

**Show Your work!**

c. How does boron's atomic mass compare with the mass numbers of its isotopes?

2.) A nitrate of lead is found to contain 45.52% lead. What is the formula for this salt?

3.) A hydrocarbon with a molar mass of approximately 85, is found to contain 85.7% carbon by mass.

a.) calculate its empirical formula:

b.) Determine its molecular formula:

4) Calculate number of oxygen atoms in 20.00g of zinc phosphate.

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**Princeton Review Staff, The Princeton  
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