

**Question:** How does light reveal the behavior of electrons in an atom?

From fireworks to stars, the color of light is useful in finding out what's in matter. The emission of light by hydrogen and other atoms has played a key role in understanding the electronic structure of atoms. Trace materials, such as evidence from a crime scene, lead in paint or mercury in drinking water, can be identified by heating or burning the materials and examining the color(s) of light given off in the form of bright-line spectra.

### Model 1 – White Light



- Trace the arrows in Model 1 with the appropriate color and highlight in the table using Notability.
- What happens to white light when it passes through a prism? (please type your answer)  
It splits up and the colors separate through dispersion.
- Why are the color labels in the table in Model 1 plural (i.e., "Reds" rather than "Red")? (please type your answer)  
The frequencies fall between the colors and can be produced in different shades.
- Do **all colors** of light travel at the same speed? (please type your answer)  
All colors travel at the same speeds.
- Energy in chemistry is measured in joules which is represented by the letter J. Do all colors of light have the same energy? If no, which colors have the highest energy and the least energy, respectively? (please type your answer)  
No, colors of light differ in energy. Violet has the highest energy and red has the least.

# Ap Chem Electron Energy And Light Pogil Answers

**Christian Drosten**



**Ap Chem Electron Energy And Light Pogil Answers:**

**Science Citation Index** ,1992 Vols for 1964 have guides and journal lists      Photo-electricity Herbert Stanley Allen,1913      *Electron Energy Levels* ,1962      The Energy of Photo-electrons from Sodium and Potassium as a Function of the Frequency of the Incident Light William Henry Kadesch,1914      *Electron Energy Levels in Ordered and Disordered Structures* Marilyn E. Usher,1994

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