

# INORGANIC CHEMISTRY II ASSIGNMENT

## Question 1

Chemical species	oxidation state of chromium	color of compound
$\text{Cr}^{3+}$	+3	purple/violet
$\text{CrO}_4^{2-}$	+6	yellow
$\text{Cr}_2\text{O}_7^{2-}$	+6	orange/red-orange
$\text{CrO}_2$	+4	dark blue-black
$\text{Cr}_2\text{O}_3$	+3	green
$\text{Cr}_2\text{S}_3$	+3	purple-black

1a)



1b) 100 g 90%

100 g

100 g

100 g

100 g

100 g 90% = 90 g 100 g 100% = 100 g

100 g 100%

1c)

Chemical species	oxidation state of chromium	color
$\text{Cr}^{3+}$	+3	purple
$\text{CrO}_4^{2-}$	+6	yellow
$\text{Cr}_2\text{O}_7^{2-}$	+6	orange
$\text{Cr}_2\text{S}_3$	+3	dark blue-black



## Question 2



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**J. A. Buckley, F. W. Elmer**



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Duward Shriver, Mark Weller, Tina Overton, Fraser Armstrong, Jonathan Rourke, Dynamic Books, 2016-03-25

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