

Three Fundamental Elements of Chemical Reaction

Chemical Reaction

Reaction: any process in which substances (reactants) combine to form new substances (products) with different chemical and physical properties.

Chemical reactions are classified into three types: combination, decomposition, and displacement. In this section, we will discuss the first two types of chemical reactions.

Reaction	Chemical Equation	Word Equation
Combination	$2H_2 + O_2 \rightarrow 2H_2O$	Hydrogen + Oxygen \rightarrow Water
Decomposition	$2H_2O \rightarrow 2H_2 + O_2$	Water \rightarrow Hydrogen + Oxygen
Displacement	$Zn + CuSO_4 \rightarrow ZnSO_4 + Cu$	Zinc + Copper Sulfate \rightarrow Zinc Sulfate + Copper
Double Displacement	$NaCl + AgNO_3 \rightarrow NaNO_3 + AgCl$	Sodium Chloride + Silver Nitrate \rightarrow Sodium Nitrate + Silver Chloride

Chemical reactions are classified into three types: combination, decomposition, and displacement. In this section, we will discuss the first two types of chemical reactions.

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