

# Chemical Reactions

## 40 Sample Test Questions

### Multiple Choice

Identify the choice that best completes the statement or answers the question.

- Chemical reactions \_\_\_\_\_.  
A) occur only in living organisms  
B) create and destroy atoms  
C) only occur outside living organisms  
D) produce new substances
- Everyday equations describe \_\_\_\_\_.  
A) thermonuclear reactions  
B) everyday processes  
C) chemical reactions  
D) biological chemistry
- Chemical equations \_\_\_\_\_.  
A) describe chemical reactions  
B) show how to write chemical formulas  
C) give directions for naming chemical compounds  
D) describe only biological changes
- Symbols used in equations, together with the explanations of the symbols, are shown below. Which set is correct?  
A) (g), grams  
B) (l), liters  
C) (aq), dissolved in water  
D) (s), solid product
- Which of the following is the correct skeleton equation for the reaction that takes place when solid phosphorus combines with oxygen gas to form diphosphorus pentoxide?  
A)  $P(s) + O_2(g) \rightarrow PO_2(g)$   
B)  $P(s) + O(g) \rightarrow P_2O_5(g)$   
C)  $P(s) + O_2(g) \rightarrow P_2O_5(s)$   
D)  $P_2O_5(s) \rightarrow P_2(s) + O_2(g)$
- If you rewrite the following word equation as a balanced chemical equation, what will the coefficient and symbol for fluorine be?  
nitrogen trifluoride  $\rightarrow$  nitrogen + fluorine  
A)  $6F_2$   
B)  $F_2$   
C)  $6F$   
D)  $3F_2$
- What are the coefficients that will balance the skeleton equation below?  
 $AlCl_3 + NaOH \rightarrow Al(OH)_3 + NaCl$   
A) 1, 3, 1, 3  
B) 3, 1, 3, 1  
C) 1, 1, 1, 3  
D) 1, 3, 3, 1
- When the equation  $Fe + Cl_2 \rightarrow FeCl_3$  is balanced, what is the coefficient for  $Cl_2$ ?  
A) 1  
B) 2  
C) 3  
D) 4
- Chemical equations must be balanced to satisfy \_\_\_\_\_.  
A) the law of definite proportions  
B) the law of multiple proportions  
C) the law of conservation of mass  
D) Avogadro's principle
- When the equation  $KClO_3(s) \rightarrow KCl(s) + O_2(g)$  is balanced, the coefficient of  $KClO_3$  is \_\_\_\_\_.  
A) 1  
B) 2  
C) 3  
D) 4
- The product of a combination reaction is  $Ba(OH)_2$ . If one of the reactants is  $H_2O$ , what is the other reactant?  
A)  $Ba_2O$   
B)  $BaO$   
C)  $BaH$   
D)  $BaO_2$
- In order for the reaction  $2Al + 6HCl \rightarrow 2AlCl_3 + 3H_2$  to occur, which of the following must be true?  
A) Al must be above Cl on the activity series.  
B) Al must be above H on the activity series.  
C) Heat must be supplied for the reaction.  
D) A precipitate must be formed.

# Answers To Test A 11 Chemical Reactions

**Arshad Iqbal**



## **Answers To Test A 11 Chemical Reactions:**

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