

Unit 14 – Acid, Bases, & Salts

The pH Scale

- Hydronium & Hydroxide Ions
(H_3O^+) (OH^-)
- pH Scale
- Calculations



Answers To Unit 14 Acids And Bases

A Gutmann



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are proton donor compounds that are measurably ionized in water such as aliphatic and aromatic carboxylic acids substituted phenols and mineral acids The bases of principal interest are likewise compounds capable of forming ions in water for example aliphatic and aromatic amines and derivatives of guanidine or pyridine The solvents emphasized are hydrocarbons and halohydrocarbons but data for dipolar aprotic solvents for example acetone acetonitrile and nitrobenzene are included Contrasts in acid base behavior and in acidity and basicity scales in aprotic and water like solvents are discussed The role of hydrogen bonding in aprotic solvents is discussed at length Important types of hydrogen bonded structures include chelate rings self associated acids bases and salts hydrogen bonded ion pairs and homo and heteroconjugate cations and anions Examples are given in which hydrogen bonding of these types affects such properties as the absorption spectrum of a salt the catalytic effect of an acid and the accurate location of a titration endpoint Author **AP Chemistry Premium, 2026: Prep**

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