#### AP Biology Study Guide Unit 1

#### Properties of Water:

- 1. Polarity allows H20 to form Hydrogen Bonds
- 2. Cohesion Water H bonds to itself, creating surface tension
- 3. Adhesion Water adheres to other polar substances
  - a. Allows easy transport of water against Gravity
- 4. Universal Solvent Water can dissolve Polar and Ionic Compounds
  - a. This allows for reactions between Aqueous Solutions
- 5. Expands when Frozen this means that Ice floats
- High Specific Heat water can absorb or release a large amount of heat with only a slight change in its own temperature (caused by cohesion)
  - High Heat of Vaporization only warmest molecules can evaporate, so this causes evaporative cooling
- 7. Can be regulated by non-polar hydrophobic substances
- Neutral PH → can form acids or bases if substances are added

Polarity - Difference in electronegativity between atoms bonded together create slightly charged ends of the molecules they form

Hydrophobic Substance - Nonpolar substance that repels water Hydrophilic Substance - Polar substance that attracts water molecules

#### CHON - Carbon, Hydrogen, Nitrogen, Oxygen

- Makeup 99% of living matter
- Strong Biological Cornerstones Because:
  - Smallest Diameter/Valency for atoms of 1, 2, 3, 4 valence electrons
  - Form Stable Covalent Bonds
  - Can form double and triple bonds
  - Versatile

#### Biological Transition Metals - Mn. Fe. Co. Cu. Zn.

Can exist in multiple oxidation states → can be both oxidized and reduced

#### Biological Ions: Na+, Mg2+, Cl-, K+, Ka2+

Used to maintain homeostasis and osmotic concentration

Phosphorous/Sulphur - Energy Carriers which form strong high energy bonds

#### 5 Foundation Biological Molecules:

# **36 Guide Ap Biology**

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# **36 Guide Ap Biology:**

AP Biology Study Guide Sundar Nathan, 2009-11 Sundar Nathan received a Bachelor's degree in Electrical Engineering from Anna University Chennai India and a Masters degree in Biomedical Engineering from the University of Texas at Austin Working for over a year with a team of talented Phds MPhils and MScs from all over the world Sundar compiled this comprehensive study guide to help students prepare diligently understand the concepts and Crush the AP Bio Test

Learning and Understanding National Research Council, Division of Behavioral and Social Sciences and Education, Center for Education, Committee on Programs for Advanced Study of Mathematics and Science in American High Schools, 2002-09-06 This book takes a fresh look at programs for advanced studies for high school students in the United States with a particular focus on the Advanced Placement and the International Baccalaureate programs and asks how advanced studies can be significantly improved in general It also examines two of the core issues surrounding these programs they can have a profound impact on other components of the education system and participation in the programs has become key to admission at selective institutions of higher education By looking at what could enhance the quality of high school advanced study programs as well as what precedes and comes after these programs this report provides teachers parents curriculum developers administrators college science and mathematics faculty and the educational research community with a detailed assessment that can be used to guide change within advanced study programs Readers' Guide to Periodical Literature Anna Lorraine Guthrie, 1915 An author subject index to selected general interest periodicals of reference value in libraries Guide to College Majors 2009 Princeton Review, 2009 Provides information on more than four hundred undergraduate majors including related fields sample college curricula suggested high school preparation courses and career and salary prospects for graduates Reader's Guide to Periodical Literature Supplement ,1928

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