

AAMC Practice Test 2 exam 2023/2024 with 100% correct answers

Chem and Physics section - correct answer

What atom is the site of covalent attachment of AMC to the model tetrapeptide used in the studies? - correct answer Because AMC is attached to the peptide on the carboxyl side. This suggests that an amide linkage involving the N atom in AMC is used to covalently attach the fluorophore to the peptide

*a peptide bond between amino acids= covalent bond between carboxyl group of 1 amino acid to amino group of other, releasing water

What expression gives the amount of light energy (in J per photon) that is converted to other forms between the fluorescence excitation (absorption) and emission events? - correct answer Because $E = hf = hc/\lambda$

Excitation occurs at $\lambda_{exc} = 360\text{nm}$, but fluorescence is observed at $\lambda_{em} = 440\text{nm}$. This implies that an energy of

$$E = hc \left[\left(\frac{1}{360 \times 10^{-9}} \right) - \left(\frac{1}{440 \times 10^{-9}} \right) \right]$$

$E = h \times c \times \left[\left(\frac{1}{360 \times 10^{-9}} \right) - \left(\frac{1}{440 \times 10^{-9}} \right) \right]$ J per photon converted to other forms between the excitation and fluorescence events

The concentration of enzyme for each experiment was $5.0 \mu\text{M}$. What is the k_{cat} for the reaction at pH 4.5 when no chloride added when compound 3 is substrate? - correct answer $2.5 \times 10^{-2} \text{ s}^{-1}$

Rate for compound 3 is 125 nM/s , and rate of product formation did not vary over time for first 5 minutes implies enzyme was saturated with substrate.

$$k_{cat} = V_{max}/[E] = 125 \text{ nM/s}/(5.0 \mu\text{M}) = 2.5 \times 10^{-2} \text{ s}^{-1}$$

Absorption of ultraviolet light by organic molecules always results in what process? - correct answer excitation of bound electrons (since going from low to high energy state so they're excited)

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James Oliver Nichols



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