Phase Change Worksheet

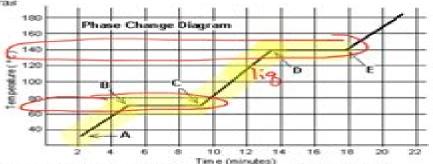
Name: Period:

The graph was drawn from data collected as a substance was heated at a constant rate.

Use the graph to answer the following questions 1-23.

Condensation

Melting deposition Sublimation



1. - 22. At point A, the beginning of observations, the substance exists in a solid state. Material in this phase has

(1) definite volume and (2) definite shape. With each passing minute, (3) heat is added to the substance. This causes the molecules of the substance to (4) Move more rapidly which we detect by a

(5) temperature is in the substance. At point B, the temperature of the substance is (6) 70 °C. The solid begins to (7) Melt. At point C, the substance is completely (8) Melted or in a (9) 1 glind state.

Material in this phase has (10) of finite volume and (11) ivd finite shape. The energy put to the substance between minutes 5 and 9 was used to convert the substance from a (12) Solid to a

(13) liquid

Between 9 and 13 minutes, the added energy increases the (14) to point E, the liquid is (15) to point E, the liquid is (15) to point E, the substance is completely in the

(16) 905 phase. Material in this phase has (17) indef volume and (18) indef shape. The

energy put to the substance between minutes 13 and 18 converted the substance from a (19) liq to a

(20) OOS state. Beyond point E, the substance is still in the (21) GOS phase, but the molecules are moving (22) as indicated by the increasing temperature.

23. Which of these three substances was likely used in this phase change experiment?

Substance	Melting point	Boiling point
Bolognium,	20.°C	100 °C
Unobtainium	40 °C	140 °C
Feesium	70 °C	140 °C

33 Phase Changes Answers

Charles Liu

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