# Practice: Balancing Equations #3

Show all work on a separate sheet of paper. Place coefficient final answers in the blanks provided. You can leave

<sup>2</sup>. \_\_\_\_MgF<sub>2</sub>+ \_\_\_\_Li<sub>2</sub>CO<sub>3</sub> 
$$\Rightarrow$$
 \_\_\_\_MgCO<sub>3</sub>+ \_\_\_\_LiF

3. \_\_\_ 
$$P_4$$
 + \_\_\_  $O_2$   $\Rightarrow$  \_\_\_  $P_2O_3$ 

4. \_\_\_\_ 
$$RbNO_3$$
 + \_\_\_\_  $BeF_2$   $\rightarrow$  \_\_\_\_  $Be(NO_3)_2$  + \_\_\_\_  $RbF$ 

5. \_\_\_\_ AgNO<sub>3</sub> + \_\_\_\_ Cu 
$$\rightarrow$$
 \_\_\_\_ Cu(NO<sub>3</sub>)<sub>2</sub> + \_\_\_\_ Ag

6. \_\_\_\_ 
$$CF_4 +$$
 \_\_\_\_  $Br_2 \Rightarrow$  \_\_\_\_  $CBr_4 +$  \_\_\_\_  $F_2$ 

7. \_\_\_\_ HCN + \_\_\_ CuSO<sub>4</sub> 
$$\Rightarrow$$
 \_\_\_\_ H<sub>2</sub>SO<sub>4</sub> + \_\_\_ Cu(CN)<sub>2</sub>

$$10.$$
\_\_\_N<sub>2</sub>+\_\_\_NH<sub>3</sub>

11. NaF + Br<sub>2</sub> 
$$\rightarrow$$
 NaBr + F<sub>2</sub>

12. 
$$Pb(OH)_2 + HCI \rightarrow H_2O + PbCl_2$$

13.\_\_\_\_ 
$$AlBr_3 +$$
 \_\_\_\_  $K_2SO_4 \rightarrow$  \_\_\_\_  $KBr +$  \_\_\_\_  $Al_2(SO_4)_3$ 

14.\_\_\_ 
$$CH_4+$$
 \_\_\_  $O_2 \rightarrow$  \_\_\_  $CO_2+$  \_\_\_  $H_2O$ 

15.\_\_\_\_ Na<sub>3</sub>PO<sub>4</sub>+\_\_\_\_ CaCl<sub>2</sub>
$$\rightarrow$$
 \_\_\_\_ NaCl + \_\_\_\_ Ca<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub>

17. 
$$AI + HCI \rightarrow H_2 + AICI_3$$

18.\_\_\_ 
$$N_2 +$$
\_\_\_  $F_2 \rightarrow$ \_\_\_  $NF_3$ 

19.\_\_\_ 
$$SO_2 +$$
 \_\_\_  $Li_2Se \rightarrow$  \_\_\_  $SSe_2 +$  \_\_\_  $Li_2O$ 

20. 
$$NH_3 + H_2SO_4 \Rightarrow (NH_4)_2SO_4$$

# **9 Reactions And Equations Answers**

**Charles Henrickson** 

#### **9 Reactions And Equations Answers:**

A New Approach to ICSE Chemistry for Class IX (A.Y. 2023-24)Onward V. K. Sally, 2023-05-20 The book A New Approach to I C S E Chemistry for Class IX has been written in accordance with the latest syllabus prescribed by the Council for Indian School Certificate Examination New Delhi The book is divided into five distinct parts 1 The first part constitutes chapters on The Languages of chemistry Chemical Changes and Reactions and Water to give an idea to the students about the diversity of matter 2 The second part constitutes chapters on Atomic Structure and Chemical Bonding and The Periodic Table This part will help the students to understand unity in diversity of matter It further explains the rules required for reaction between various forms of matter 3 The third part constitutes chapter on Study of the First Element Hydrogen and Study of Gas Laws This will familiarise the students with element hydrogen and their chemical properties It further explains the behaviour of gases under changes of temperature and pressure and also provides explanation in terms of molecular motion 4 The fourth part deals with Atmospheric Pollution 5 The fifth part deals with Practical Chemistry The salient features of the book Large number of chemical reactions are described with experimental observations Important points have been highlighted for location of precise answers All definitions and other important points are highlighted Diagrams are made simple and more Informative In the chapter on the language of chemistry a new method has been introduced for balancing chemical equations This method Is based on atomic numbers which Is fundamental property of the elements A chapter on Practical Chemistry is added to help students for laboratory work Questions have been added chapter wise under the heading Question Bank for the benefit of students The solution can be accessed through QR Code given at the end of each chapter Periodic Table showing mass number atomic number of various elements along with list of actual names of the elements is provided A Specimen Paper Solved has also been included through QR Code for the benefit of students Annual Examination Paper 2019 Solved has been given through OR Code Model Test Papers 1 to 5 Unsolved have been given through OR Code We hope this book will prove very useful to the students and teachers Suggestions and constructive criticism for the further improvement of the book would be gratefully acknowledged Author **Survival Guide to General Chemistry** Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics The focus is not to recap or review the theoretical concepts well described in the available texts Instead the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester Each is written in a specific and detailed step by step process for problem solving

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