

CHEM 111 Study Guide for Exam 4: Chapters 7–9

Chapter 7: Chemical reactions and chemical quantities

1. Know how to do gram / mole conversions.
2. Know the relationships between moles and chemical equations, and how to do calculations based on a balanced equation
3. Know the relationships between mass and chemical equations, and how to do calculations based on a balanced equation
4. Be able to figure out limiting reagent and be able to calculate theoretical yield and percent yield.

Chapter 8: Introduction to Solutions and Aqueous Reactions

1. Understand molarity as a concentration and be able to use it in calculations.
2. Be able to do dilution stoichiometry.
3. Know the difference among strong electrolytes, weak electrolytes and nonelectrolytes.
4. Be able to do solution stoichiometry in a given chemical reaction.
5. Be able to write molecular equations, complete ionic equations and net ionic equations.
6. Be able to assign oxidation numbers (states) for a given chemical formula.
7. Be able to recognize redox reactions, and identify oxidation reaction, reduction reaction, oxidizing agent and reducing agent.

Chapter 9: Thermochemistry

1. Know the difference of kinetic energy and potential energy.
2. Understand endothermic reactions and exothermic reactions.
3. Be able to write equations for standard enthalpy of formation (ΔH_f°).
4. Be able to use Hess's law.
5. Be able to use bond energies to calculate ΔH for a chemical reaction.
6. Be able to use given ΔH_f° to calculate $\Delta H_{\text{rxn}}^\circ$
7. Be able to calculate heat (q) using specific heat: $q = c.m. \Delta T$

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Occupational Safety and Hygiene III Pedro M. Arezes, João Santos Baptista, Monica P. Barroso, Paula Carneiro, Patrício Cordeiro, Nelson Costa, Rui B. Melo, A. Sergio Miguel, Gonçalo Perestrelo, 2015-02-02 The papers published in Occupational Safety and Hygiene III cover the following topics Occupational safety Risk assessment Safety management Ergonomics Management systems Environmental ergonomics Physical environments Construction safety and Human factors The contributions are based on research carried out at universities and other resea

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