# CHEM 121 ACS FINAL MOCK EXAM KEY JUNE 2023 WITH 100% ACCURATE ANSWERS (HIGHLY

## RECOMMEDED)

- 1. Which pair of compounds will form hydrogen bonds with one another?
  - a. CH<sub>e</sub> and H<sub>e</sub>O
  - b. CH, and NH,
  - c. HF and CH.
  - d. H<sub>2</sub>O and NH<sub>2</sub>
- 2. Which substance is most soluble in water?
  - a. ethane, CH<sub>2</sub>CH<sub>3</sub>
  - b. ethanol, CH3CH-OH
  - c. n-butane, CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>CH<sub>3</sub>
  - I-butanol, CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>OH
- 3. A 250-mL bottle of a sports drink solution contains 4.50% by mass of sodium chloride. What is the molal concentration of sodium chloride in this bottle of sports drink?
  - 40.3006 au
  - b. 0.770 mm
  - c. 0.180 m
  - d. 0.0113.m
- 4. What mass of water is needed to dissolve 292.5 g of NaCl to produce a 0.25 m aqueous solution?
  - A 20 kg
  - b. 5.0 kg
  - c. 0.80 kg
  - d. 0.0650 kg
- 5. A 2.50 g sample of naphthalene, C<sub>n</sub>H<sub>n</sub>, was dissolved in 100 g of benzene. What is the freezing point of the benzene solution? The freezing point of pure benzene is 5.45 °C; K<sub>t</sub> = 5.07 °C ·m<sup>-1</sup>.
  - a. -0.989 °C
  - ы. 0.989 °С

Molar mass / g-mol <sup>4</sup> C.-H. 128 17

Modar mass / g-mod \*

# **Acs Chemistry 121 Exam**

**Arturo Cuomo** 

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