

1. According to the Bronsted—Lowry definition, which chemical species can function *both* as an acid and as a base?

- Cl^-
- SO_4^{2-}
- NH_4^+
- HCO_3^-
- H_3O^+

2. In the reaction $\text{CN}^- + \text{H}_2\text{O} \rightleftharpoons \text{HCN} + \text{OH}^-$ which is an acid—base conjugate pair?

- H_2O and HCN
- H_2O and OH^-
- CN^- and H_2O
- HCN and OH^-

3. Given that HX is a stronger Bronsted acid than HY in aqueous solution, which is true of a 1 M solution of NaX ?

- It is less basic than a 1 M solution of NaY .
- It is more basic than a 1 M solution of NaY .
- It yields a neutral solution.
- It is more concentrated than a 1 M solution of NaY .

4. The pH of a 0.03 M HCl solution is

- 1.5
- 2.5
- 3.5
- 12.5

5. The pH of a 1.0×10^{-2} M $\text{Ba}(\text{OH})_2$ solution at 25°C is

- 2.7
- 3.0
- 11.0
- 11.3

6. Which is the strongest acid?

- HClO
- HClO_2
- HClO_3
- HClO_4

7. Which salt reacts with water (hydrolyzes) to produce a basic solution?

- $\text{NaC}_2\text{H}_3\text{O}_2$
- NaNO_3
- NH_4Cl
- BaSO_4

8. The *weakest* of the bases listed is

Acid	Conjugate Base	K_a (Ionization Constant of Acid)
HCl	Cl^-	100% ionized
HSO_4^-	SO_4^{2-}	1.2×10^{-2}
H_2S	HS^-	5.7×10^{-8}
HS^-	S^{2-}	1.2×10^{-13}

- Cl^-
- CN^-
- HS^-
- S^{2-}
- SO_4^{2-}

9. The oxide of which element will react with water to form the strongest acid?

	Main Groups							
Period	I	II	III	IV	V	VI	VII	O
First								
Second	X	Y	Z	P		Q	S	U
Third	W					R	T	M

- W
- M
- P
- R
- Z

10. Which statement is a logical inference from the fact that a 0.10 M solution of potassium acetate $\text{KC}_2\text{H}_3\text{O}_2$ is less alkaline than a 0.10 M solution of potassium cyanide, KCN ?

- Hydrocyanic acid is a weaker acid than acetic acid.
- Hydrocyanic acid is less soluble in water than acetic acid.
- Cyanides are less soluble than acetates.
- Acetic acid is a weaker acid than hydrocyanic acid.

11. In the titration of 50.0 mL of 0.100 M benzoic acid (a monoprotic acid) with 50.0 mL of 0.100 M NaOH , the properties of the solution at the equivalence point will correspond exactly to the properties of

- a 0.100 M sodium solution.
- a 0.0500 M sodium hydroxide solution.
- a 0.0500 M benzoic acid solution.
- a 0.0500 M sodium benzoate solution.

Acs Practice Test 2 Answers

William Watson (of Preston.)



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CCSP CSI Exam Cram 2 (Exam Cram 642-541) Annlee Hines, 2003 Annotation In this book you ll learn how to identify the information assets in a network which must be protected recognize which threats apply to which network assets choose an appropriate technology to mitigate a threat apply network defense in layers choose the appropriate security product to protect the network ingress from the Internet DMZ VPNs campus servers management network branches protect network devices from misuse by hackers and choose alternative security designs BOOK JACKET Title Summary field provided by Blackwell North America Inc All Rights Reserved **Cracking the SAT French Subject Test, 2009-2010 Edition**

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The new code, 1885. The quarterly arithmetic. Standard i(-vii). [With] Answers. Standard i,iv(-vii). William Watson (of Preston.),1893 Field Manuals United States. War Department,1948 *Cracking the AP World History Exam, 2013 Edition* Princeton Review,2012-08-07 Provides test taking strategies key terms a subject review and two full length practice tests Dr Sunil's One Page Solutions for General Practice K Sunil Ravinder Paul,2022-12-20 1 General Topics Common Diseases Problems 2 Pain 3 Emergencies 4 Trauma Advanced Trauma Life Support 5 Vaccination Immunization for Children and Adults 6 Systems and Organs 7 Cancer 8 Geriatrics 9 Types of Anesthesia 10 Needles Tubes and Drains 11 Microbes and Relevance to Drugs 12 Drugs 13 Intravenous Fluids and Electrolytes 14 Nutrition 15 Sterilization and Disinfection For Clinics Wards Operation Theater 16 Helpful Hints for Doctors 17 ABCs for Doctors and How to Avoid Litigations 18 Investigations 19 Postoperative Complications Complications for Specific Surgery 20 Prescription Writing

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take the first step and develop a solid research question As the field of language education evolves we need continual research to improve our instructional and assessment practices and our understanding of the learners language learning processes This book with its remarkable 150 topics and 10 times the number of potential research questions provides a wealth of ideas that will help early career researchers conduct studies that move our field forward and grow our knowledge base Deborah J Short Ph D Director Academic Language Research Training Past President TESOL International Association 2021 22 As a teacher in graduate programs in TESOL I frequently come across the frustration of students at centering their research interests on a particular topic and developing research questions which are worth pursuing so as to make a contribution to the field This frustration stems from the fact that our field is so vast and interrelated that it is often impossible to properly address all that interests them Hence I wholeheartedly welcome this most relevant and innovative addition to the research literature in the field of TESOL and Applied Linguistics Coombe and Mohebbi have created a real tour de force that stands to inform budding researchers in the field for many years to come Additionally the cutting edge depiction of the field and all it has to offer will no doubt update the research agendas of many seasoned researchers around the world The 150 chapters are organized in a most powerful yet deceptively simple way offering a positioning within the topic suggesting questions that might direct inquiry and offering a basic set of bibliographic tools to start the reader in the path towards research What is more the nine sections in which the chapters are organized leave no area of the field unexplored Dr Gabriel D az Maggioli Academic Advisor Institute of Education Universidad ORT del Uruguay President IATEFL Chapter Metacognition in Academic Writing Learning Dimensions is available open access under a Creative Commons Attribution 4 0 International License via link [springer.com](https://www.springer.com)

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