# ACS ORGANIC CHEMISTRY FINAL STUDY GUIDE

- 1. no pi bonds (unsaturation): 0 degrees
- 1 pi bond (unsaturation): 1 degree
- triple bond (unsaturation): 2 degrees
- formal charge=: # valence e's (# nonbonding e's + 1/2 # bonding e's)
- 4 regions of e- density: sp3 hybridization, 109.5 bond angle
- 6. 3 regions of e-density: sp2 hybridization, 120 bond angle
- 2 regions of e- density: sp hybridization, 180 bond angle
- aliphatic: Pertaining to any member of one of the two major groups of organic compounds, those having straight or branched chain structures.
- 9. vinylic: the carbon atoms of a carbon-carbon double bond
- 10. higher stability=: lower energy
- 11. higher acidity=: higher conjugate base stability
- 12. higher acidity=: lower basicity
- higher pKa=: stronger base
- 14. Zaitsev's Rule: The most substituted alkene is formed preferentially
- 15. Markovnikov: The H atom will always bond to the C with more H
- 16. diastereomer: Stereoisomers that are not mirror images of one another
- 17. nucleophilicity: decreases across a period and increases down a group
- 18. enantiomer: nonsuperimposible mirror images
- 1680 cm-1 (IR): amide, C=O stretch
- 20. 1710 cm-1 (IR): ketone, C=O stretch
- 21. 1710 cm-1 (IR): acid, C=O stretch
- 22. 1725 cm-1 (IR): aldehyde, C=O stretch
- 23. 1735 cm-1 (IR): ester, C=O stretch
- 24. centered 3000 cm-1 (IR): acid, O-H stretch
- centered 3300 cm-1 (IR): alcohol, O-H stretch
- 26. 3100-3300 cm-1 (IR): amide/amine, N-H stretch, NHn where n=# of peaks
- 27. 2710 & 2810 cm-1 (IR): aldehyde, C-H stretch

# **Acs Organic Chemistry Final Study Guide**

**Sterling Test Prep, Frank Addivinola** 

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