- The number of σ bonds in N≡N is
 - a. 1 b. 2
 - e. 3
 - d. 4
- The elements in an ionic compound are held together by
 - electrostatic forces of attraction.
 - b. van der Waals forces.
 - e. the spin of paired electrons.
 - d. the formation of hybrid orbitals.
 - e. an electron pair.
- In every electrolytic and galvanic (voltaic) cell the anode is that electrode
 - at which exidation occurs.
 - b. which attracts cations.
 - at which electrons are supplied to the solution.
 - at which reduction occurs.
- 28. Metal X was plated from a solution containing cations of X. The passage of 48.25 C deposited 31 mg of X on the cathode. What is the mass of X (in grams) per mole of electrons?
 - a. 47
 - b. 62
 - c. 93d. 186
- 29. In a galvanic (voltaic) cell in which the reaction is Cd + Cu²⁺ → Cu + Cd²⁺ and the ions are at unit concentration (activity), the cell potential is
 - $Cd \rightarrow Cd^{2+} + 2e$ 0.4021 V $Cu \rightarrow Cu^{2+} + 2e$ 0.344 V
 - a. 0.1383 V
 - b. 0.4021 V
 c. 0.344 V
 - d. 0.7461 V
 - e. 0.3677 V
- 30. In which reaction will an increase in total pressure at constant temperature favor formation of the products?
 - a. $CaCO_2(s) \leftarrow \rightarrow CaO(s) + CO_2(g)$
 - b. $H_2(g) + Cl_2(g) \leftrightarrow 2HCl(g)$
 - c. $2NO(g) + O_2(g) \leftrightarrow 2NO_2(g)$
 - d. COCl₂(g) ←→ CO(g) + Cl₂(g)

Standard Potentials	E
$Mg \rightarrow Mg^2 + 2e$	2.37V
$Al \rightarrow Al^{5*}+3e$	1.66V
$Zn \rightarrow Zn^{2*}+2c$	0.76V
Fe⇒Fe ^{2*} +2e	0.44V
Cu → Cu ² *+2e	0.34V
Ag→ Ag'+e	-0.80V

- Using only the metals Mg, Al, Zn, Fe, Cu and Ag, together with their 1 M salt solutions, a voltaic cell of the highest possible voltage would be constructed using electrodes of these metals.
 - Mg and Ag
 - b. Mg and Fe
 - c. Zn and Cu
 - el. Al and Ag
 - e. Mg and Al
- 32. $E = E^{\circ} 0.059/n \log Q$ (Nernst equation) $[H^{\circ}] = 1.0 \text{ M initially}, P_{o2} = 1.0 \text{ atm}$

4c + O₂(g)+4H*(aq)←→ 2H₂O(l) E*=1.23V Based on the information above, which statement is correct?

- a. n = 1, since one mole of oxygen is being considered.
- Addition of base should result in an E value, which is less than 1.23 V.
- c. E is independent of the pH of the solution.
- d. $Q = [H_2O]^2$ $[O_2][H^*]$
- 33. The equilibrium constant for the gaseous reaction C + D ← → E + 2F is 3.0 at 50 °C. In a 2.0 L flask at 50 °C are placed 1.0 mol of C, 1.0 mol of D, 1.0 mol of E, and 3.0 mol of F. Initially, the reaction will
 - a. proceed at equal rates in both directions.
 - b. proceed more rapidly to form E and F.
 - proceed more rapidly to form C and D.
 - not occur in either direction.

Compound	ΔG° _ε kJ/mol	
H ₂ O(l)	-237	
H ₂ O(g)	-229	

- At 298 K the equilibrium constant for H₂(g) + ½ O₂(g) ← → H₂O(I)
 - is larger than the Keq for H₂(g) + ½ O₂(g) ← → H₂O(g)
 - b. will have a value of 1 0 at equilibrium.
 - c. cannot be computed since data on O2 and

Acs College Chemistry Exam 2011

Julee B. Waldrop, Melody A. Bowdon

Acs College Chemistry Exam 2011:

Undergraduate Chemistry Education National Research Council, Division on Earth and Life Studies, Board on Chemical Sciences and Technology, Chemical Sciences Roundtable, 2014-03-24 Undergraduate Chemistry Education is the summary of a workshop convened in May 2013 by the Chemical Science Roundtable of the National Research Council to explore the current state of undergraduate chemistry education Research and innovation in undergraduate chemistry education has been done for many years and one goal of this workshop was to assist in the transfer of lessons learned from the education research community to faculty members whose expertise lies in the field of chemistry rather than in education Through formal presentations and panel discussions participants from academia industry and funding organizations explored drivers of change in science technology engineering and mathematics education innovations in chemistry education and challenges and opportunities in chemistry education reform Undergraduate Chemistry Education discusses large scale innovations that are transferable widely applicable and or proven successful with specific consideration of drivers and metrics of change barriers to implementation of changes and examples of innovation in the classroom Forensic Chemistry Kelly M. Elkins, 2018-09-03 Chemistry Forensic Science Forensic chemistry is a subdiscipline of forensic science its principles guide the analyses performed in modern forensic laboratories Forensic chemistry s roots lie in medico legal investigation toxicology and microscopy and have since led the development of modern forensic analytic techniques and practices for use in a variety of applications Introduction to Forensic Chemistry is the perfect balance of testing methods and application Unlike other competing books on the market coverage is neither too simplistic nor overly advanced making the book ideal for use in both undergraduate and graduate courses The book introduces chemical tests spectroscopy advanced spectroscopy and chromatography to students The second half of the book addresses applications and methods to analyze and interpret controlled substances trace evidence questioned documents firearms explosives environmental contaminants toxins and other topics The book looks at innovations in the field over time including the latest development of new discernible chemical reactions instrumental tools methods and more Key features Nearly 300 full color figures illustrating key concepts and over 20 case studies Addresses all the essential topics without extraneous or overly advanced coverage Includes full pedagogy of chapter objectives key terms lab problems end of chapter questions and additional readings to emphasize key learning points Includes chemical structures and useful spectra as examples Fulfils the forensic chemistry course requirement in FEPAC accredited programs Includes a chapter on Chemical Biological Radiological Nuclear and Explosive CBRNE materials Comprehensive and accessible without being overly technical Introduction to Forensic Chemistry will be a welcome addition to the field and an ideal text designed for both the student user and professor in mind Course ancillaries including an Instructor's Manual with Test Bank and chapter PowerPoint lecture slides are available with qualified course adoption Best Practices for Flipping the College Classroom Julee B.

Waldrop, Melody A. Bowdon, 2015-06-26 Best Practices for Flipping the College Classroom provides a comprehensive overview and systematic assessment of the flipped classroom methodology in higher education The book Reviews various pedagogical theories that inform flipped classroom practice and provides a brief history from its inception in K 12 to its implementation in higher education Offers well developed and instructive case studies chronicling the implementation of flipped strategies across a broad spectrum of academic disciplines physical environments and student populations Provides insights and suggestions to instructors in higher education for the implementation of flipped strategies in their own courses by offering reflections on learning outcomes and student success in flipped classrooms compared with those employing more traditional models and by describing relevant technologies Discusses observations and analyses of student perceptions of flipping the classroom as well as student practices and behaviors particular to flipped classroom models Illuminates several research models and approaches for use and modification by teacher scholars interested in building on this research on their own campuses The evidence presented on the flipped classroom methodology by its supporters and detractors at all levels has thus far been almost entirely anecdotal or otherwise unreliable Best Practices for Flipping the College Classroom is the first book to provide faculty members nuanced qualitative and quantitative evidence that both supports and challenges the value of flipping the college classroom Challenges in Chemistry Graduate Education National Research Council, Division on Earth and Life Studies, Board on Chemical Sciences and Technology, Committee on Challenges in Chemistry Graduate Education; Board on Chemical Sciences, 2012-08-21 Chemistry graduate education is under considerable pressure Pharmaceutical companies long a major employer of synthetic organic chemists are drastically paring back their research divisions to reduce costs Chemical companies are opening new research and development facilities in Asia rather than in the United States to take advantage of growing markets and trained workforces there Universities especially public universities are under significant fiscal constraints that threaten their ability to hire new faculty members Future federal funding of chemical research may be limited as the federal budget tightens All of these trends have major consequences for the education of chemistry graduate students in U S universities To explore and respond to these intensifying pressures the Board on Chemical Sciences and Technology held a workshop in Washington DC on January 23 24 2012 titled Graduate Education in Chemistry in the Context of a Changing Environment The workshop brought together representatives from across the chemical enterprise representing leaders and future leaders of academia industry and government The goal of the workshop was not to come to conclusions but to have an open and frank discussion about critical issues affecting chemistry graduate education such as the attraction and retainment of the most able students to graduate education financial stressors on the current support model and their implications for the future model competencies needed in the changing job market for Ph D chemists and competencies needed to address societal problems such as energy and sustainability Challenges in Chemistry Graduate Education A Workshop Summary is organized into six chapters and summarizes the workshop on

Graduate Education in Chemistry in the Context of a Changing Environment POGIL Shawn R. Simonson, 2023-07-03 Process Oriented Guided Inquiry Learning POGIL is a pedagogy that is based on research on how people learn and has been shown to lead to better student outcomes in many contexts and in a variety of academic disciplines Beyond facilitating students mastery of a discipline it promotes vital educational outcomes such as communication skills and critical thinking Its active international community of practitioners provides accessible educational development and support for anyone developing related courses Having started as a process developed by a group of chemistry professors focused on helping their students better grasp the concepts of general chemistry. The POGIL Project has grown into a dynamic organization of committed instructors who help each other transform classrooms and improve student success develop curricular materials to assist this process conduct research expanding what is known about learning and teaching and provide professional development and collegiality from elementary teachers to college professors As a pedagogy it has been shown to be effective in a variety of content areas and at different educational levels. This is an introduction to the process and the community Every POGIL classroom is different and is a reflection of the uniqueness of the particular context the institution department physical space student body and instructor but follows a common structure in which students work cooperatively in self managed small groups of three or four The group work is focused on activities that are carefully designed and scaffolded to enable students to develop important concepts or to deepen and refine their understanding of those ideas or concepts for themselves based entirely on data provided in class not on prior reading of the textbook or other introduction to the topic The learning environment is structured to support the development of process skills such as teamwork effective communication information processing problem solving and critical thinking The instructor's role is to facilitate the development of student concepts and process skills not to simply deliver content to the students The first part of this book introduces the theoretical and philosophical foundations of POGIL pedagogy and summarizes the literature demonstrating its efficacy The second part of the book focusses on implementing POGIL covering the formation and effective management of student teams offering guidance on the selection and writing of POGIL activities as well as on facilitation teaching large classes and assessment The book concludes with examples of implementation in STEM and non STEM disciplines as well as guidance on how to get started Appendices provide additional resources and information about The POGIL Project Teaching Innovation in University Education: Case Studies and Main Practices Saura, Jose Ramon, 2022-06-17 In the last decade the development of new technologies has made innovation a fundamental pillar of education Teaching innovation includes the evolution of both teaching and learning models to drive improvements in educational methodologies Teaching innovation is a pioneer in the understanding and comprehension of the different teaching methodologies and models developed in the academic area Teaching innovation is a process that seeks validation in the academic and teaching communities at universities in order to promote the improvement and its practices and uses in the future characterized by digital development and data based

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The Strength in Numbers Barry Bozeman, Jan Youtie, 2020-07-14 Why collaborations in STEM fields succeed or fail and how to ensure success Once upon a time it was the lone scientist who achieved brilliant breakthroughs No longer Today science is done in teams of as many as hundreds of researchers who may be scattered across continents These collaborations can be powerful but they also demand new ways of thinking The Strength in Numbers illuminates the nascent science of team science by synthesizing the results of the most far reaching study to date on collaboration among university scientists Drawing on a national survey with responses from researchers at more than one hundred universities archival data and extensive interviews with scientists and engineers in over a dozen STEM disciplines Barry Bozeman and Jan Youtie establish a framework for characterizing different collaborations and their outcomes and lay out what they have found to be the gold standard approach consultative collaboration management The Strength in Numbers is an indispensable guide for scientists interested in maximizing collaborative success

Eye Tracking for STEM Education Research: New Perspectives Pascal Klein, Martin Rusek, Maike Schindler, 2024-04-04 A modern approach to improving education uses the components of experimental scientific research practices based on objective data dissemination of results and the use of modern technologies STEM education research is maturing and new tools and analysis techniques become available As one example eye tracking the recording of persons eye movements has been growing in popularity as it enables researchers to study

learning materials effectiveness problem solving and even students approaches during experimentation Eye movements as captured using eye tracking can reveal information about a student's attention and cognition on a process level going well beyond classical product based assessment techniques such as questionnaires or tests *Inorganic Metal Oxide Nanocrystal* Photocatalysts for Solar Fuel Generation from Water Troy K. Townsend, 2014-03-19 Troy Townsend s thesis explores the structure energetics and activity of three inorganic nanocrystal photocatalysts The goal of this work is to investigate the potential of metal oxide nanocrystals for application in photocatalytic water splitting which could one day provide us with clean hydrogen fuel derived from water and solar energy Specifically Townsend s work addresses the effects of co catalyst addition to niobium oxide nanotubes for photocatalytic water reduction to hydrogen and the first use of iron oxide rust in nanocrystal suspensions for oxygen production In addition Townsend studies a nickel oxide strontium titanate nanocomposite which can be described as one of only four nanoscale water splitting photocatalysts. He also examines the charge transport for this system Overall this collection of studies brings relevance to the design of inorganic nanomaterials for photocatalytic water splitting while introducing new directions for solar energy conversion Multiphase Catalytic Reactors Zeynep Ilsen Önsan, Ahmet Kerim Avci, 2016-06-09 Provides a holistic approach to multiphase catalytic reactors from their modeling and design to their applications in industrial manufacturing of chemicals Covers theoretical aspects and examples of fixed bed fluidized bed trickle bed slurry monolith and microchannel reactors Includes chapters covering experimental techniques and practical guidelines for lab scale testing of multiphase reactors Includes mathematical content focused on design equations and empirical relationships characterizing different multiphase reactor types together with an assortment of computational tools Involves detailed coverage of multiphase reactor applications such as Fischer Tropsch synthesis fuel processing for fuel cells hydrotreating of oil fractions and biofuels processing Why We Teach Science John L. Rudolph, 2022-12-20 Few people question the importance of science education in American schooling The public readily accepts that it is the key to economic growth through innovation develops the ability to reason more effectively and enables us to solve the everyday problems we encounter through knowing how the world works Good science teaching results in all these benefits and more or so we think But what if all this is simply wrong What if the benefits we assume science education produces turn out to be an illusion nothing more than wishful thinking In Why We Teach Science and Why We Should former high school teacher and historian of science education John L Rudolph examines the reasons we ve long given for teaching science and assesses how they hold up to what we know about what students really learn or don't learn in science classrooms and what research tells us about how people actually interact with science in their daily lives The results will surprise you Instead of more and more rigorous traditional science education to fill the STEM pipeline Rudolph challenges us to think outside the box and makes the case for an expansive science education aimed instead at rebuilding trust between science and the public something we desperately need in our current era of impending natural challenges and science denial Fullerenes—Advances in

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