

# Acids and Bases

How do acids and bases behave in water?

## Why?

Acids and bases play an important role in our lives. Numerous biological processes, industrial applications, and even environmental problems are a function of the acidity or basicity (alkalinity) of aqueous solutions. It is therefore important to understand what makes a substance behave as an acid or a base when dissolved in water. In this activity, we will explore the physical and chemical properties of acids and bases.

## Model 1 – Arrhenius Acids and Bases

Common Name for Aqueous Solution	Chemical Formula	Found in...	Tastes...	Turns Litmus Paper...	Conducts Electricity?	Acid or Base
Acetic acid	$\text{HC}_2\text{H}_3\text{O}_2(\text{aq})$	Vinegar	Sour	Red	Yes	Acid
Benzoic acid	$\text{HC}_7\text{H}_5\text{O}_2(\text{aq})$	Food preservative	Sour	Red	Yes	Acid
Phosphoric acid	$\text{H}_3\text{PO}_4(\text{aq})$	Soda pop	Sour	Red	Yes	Acid
Hydrochloric acid	$\text{HCl}(\text{aq})$	Stomach acid	Sour	Red	Yes	Acid
Citric acid	$\text{H}_2\text{C}_6\text{H}_7\text{O}_2(\text{aq})$	Citrus fruits	Sour	Red	Yes	Acid
Ascorbic acid	$\text{H}_2\text{C}_6\text{H}_7\text{O}_6(\text{aq})$	Vitamin C	Sour	Red	Yes	Acid
Magnesium hydroxide	$\text{Mg}(\text{OH})_2(\text{aq})$	Milk of magnesia	Bitter	Blue	Yes	Base
Aluminum hydroxide	$\text{Al}(\text{OH})_3(\text{aq})$	Antacids	Bitter	Blue	Yes	Base
Barium hydroxide	$\text{Ba}(\text{OH})_2(\text{aq})$	Lubricants	POISON	Blue	Yes	Base
Sodium hydroxide	$\text{NaOH}(\text{aq})$	Drain cleaner	POISON	Blue	Yes	Base

- Refer to Model 1.
  - What is the common chemical name for Vitamin C?  
**ascorbic acid**
  - Is Vitamin C classified as an acid or a base?  
**acid**
- Examine the properties of the Arrhenius acids in Model 1. List three properties that all Arrhenius acids have in common.  
**sour, red, conduct electricity**
- Examine the chemical formulas for the Arrhenius acids in Model 1. What feature do all the Arrhenius acid chemical formulas have in common?
- Examine the properties of the Arrhenius bases in Model 1. List two properties that all Arrhenius bases have in common.

# Acids And Base Answer Key Pogil

**I. M. Kolthoff**



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## **Table of Contents Acids And Base Answer Key Pogil**

1. Understanding the eBook Acids And Base Answer Key Pogil
  - The Rise of Digital Reading Acids And Base Answer Key Pogil
  - Advantages of eBooks Over Traditional Books
2. Identifying Acids And Base Answer Key Pogil
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Acids And Base Answer Key Pogil
  - User-Friendly Interface
4. Exploring eBook Recommendations from Acids And Base Answer Key Pogil
  - Personalized Recommendations
  - Acids And Base Answer Key Pogil User Reviews and Ratings
  - Acids And Base Answer Key Pogil and Bestseller Lists
5. Accessing Acids And Base Answer Key Pogil Free and Paid eBooks
  - Acids And Base Answer Key Pogil Public Domain eBooks
  - Acids And Base Answer Key Pogil eBook Subscription Services
  - Acids And Base Answer Key Pogil Budget-Friendly Options
6. Navigating Acids And Base Answer Key Pogil eBook Formats

- ePub, PDF, MOBI, and More
- Acids And Base Answer Key Pogil Compatibility with Devices
- Acids And Base Answer Key Pogil Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Acids And Base Answer Key Pogil
  - Highlighting and Note-Taking Acids And Base Answer Key Pogil
  - Interactive Elements Acids And Base Answer Key Pogil
- 8. Staying Engaged with Acids And Base Answer Key Pogil
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Acids And Base Answer Key Pogil
- 9. Balancing eBooks and Physical Books Acids And Base Answer Key Pogil
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Acids And Base Answer Key Pogil
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Acids And Base Answer Key Pogil
  - Setting Reading Goals Acids And Base Answer Key Pogil
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Acids And Base Answer Key Pogil
  - Fact-Checking eBook Content of Acids And Base Answer Key Pogil
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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