ACIDS, BASES AND SALTS

Definition of an acid	What causes the acidic properties of acids?		
An acid is a compound which when dissolved in water produces hydrogen ions as the only positively charged ions	The hydrogen ions (H') cause the acidic properties and these are formed in the presence of water.		

Another term to refer to an acid	Why an acid is also called a proton donor?	
An acid is called a <u>proton donor</u>	It's because an acid provides protons or hydrogen ions (H1) to other	
	substances during the reaction.	

Substance to which an	A base is then called	Why the base is called a	Equation for the reaction between
acid provides protons		proton acceptor?	acid and base
BASE	PROTON ACCEPTOR	Because the base accepts hydrogen ions from acids	H'mg + OHmg H ₂ O _R

Common laboratory acids		These three common laboratory acids are also called	Why these acids are also called mineral acids?	
Hydrochloric acid	HCI		They are derived from mineral salts ie chlorides for	
Sulphuric acid	H;80.	MINERAL ACIDS	HCl, sulphates for H ₂ SO ₄ and nitrates for HNO ₂	
Nitric acid	HNO ₃	STATES OF STATES		

Other minera known		Mineral salts from which the acid is derived	Organic acids known	Naturally occurring acids
Sulphurous acid	H ₂ SO ₃	Derived from sulphites	Ethanoic acid	CITRIC ACID from lemons
Carbonic acid	H ₂ CO ₂	Derived from carbonates	(CH ₃ COOH)	TARTARIC ACID from grapes
Phosphoric acid	H ₂ PO ₄	Derived from phosphates		ACETIC ACID from vinegar
Nitrous acid	HNO ₃	Derived from nitrites	Methanoic acid	LACTIC ACID from sour milk
proceeding	Marine St.		(HCOOH)	Hydrochloric acid from digestive juices

Whenever an acid is dissolved in water, it produces	Term given to the number of hydrogen ions produced by one molecule of an acid	Definition of basicity of an acid
HYDROGEN IONS	BASICITY OF AN ACID	BASICTTY of an acid is the <u>number of</u> <u>hydrogen ions</u> produced by <u>one</u> <u>molecule</u> of an acid <u>in aqueous</u> <u>solution</u> .

Acids Bases And Salts Study Guide

H Kauffman

Acids Bases And Salts Study Guide:

The book delves into Acids Bases And Salts Study Guide. Acids Bases And Salts Study Guide is a crucial topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and indepth insights into Acids Bases And Salts Study Guide, encompassing both the fundamentals and more intricate discussions.

- 1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Acids Bases And Salts Study Guide
 - Chapter 2: Essential Elements of Acids Bases And Salts Study Guide
 - Chapter 3: Acids Bases And Salts Study Guide in Everyday Life
 - Chapter 4: Acids Bases And Salts Study Guide in Specific Contexts
 - ∘ Chapter 5: Conclusion
- 2. In chapter 1, the author will provide an overview of Acids Bases And Salts Study Guide. The first chapter will explore what Acids Bases And Salts Study Guide is, why Acids Bases And Salts Study Guide is vital, and how to effectively learn about Acids Bases And Salts Study Guide.
- 3. In chapter 2, this book will delve into the foundational concepts of Acids Bases And Salts Study Guide. The second chapter will elucidate the essential principles that must be understood to grasp Acids Bases And Salts Study Guide in its entirety.
- 4. In chapter 3, the author will examine the practical applications of Acids Bases And Salts Study Guide in daily life. The third chapter will showcase real-world examples of how Acids Bases And Salts Study Guide can be effectively utilized in everyday scenarios.
- 5. In chapter 4, the author will scrutinize the relevance of Acids Bases And Salts Study Guide in specific contexts. The fourth chapter will explore how Acids Bases And Salts Study Guide is applied in specialized fields, such as education, business, and technology.
- 6. In chapter 5, this book will draw a conclusion about Acids Bases And Salts Study Guide. This chapter will summarize the key points that have been discussed throughout the book.
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