



Acids and Bases



Objectives:

- 1) What are the physical and chemical properties of acids and bases?
- 2) What criteria would you use to identify a substance as an acid, base or salt?
- 3) What are the properties of the Arrhenius, Bronsted-Lowry and Lewis models of an acid or base?
- 4) What makes an acid or base weak or strong?
- 5) What does pH and pOH mean?
- 6) How are pH and pOH determined mathematically?
- 7) How is pH and pOH related to the ion product of water?
- 8) What are the products of a neutralization reaction?
- 9) How do you write equations for a neutralization reaction?
- 10) How does the strength of an acid or base compare with the strength of its conjugate base or conjugate acid?

Assignments:

- | | Due | Grade |
|--|-------|-------|
| 1) Worksheet p. 11.4 & 11.5 (39 pts) | _____ | _____ |
| 2) p. 640 #3-4; p. 649, #19; p. 643, #9 [conjugate pairs](14 pts) | _____ | _____ |
| 3) p. 643, #10; p. 674, #107 [Lewis acids & bases](8 pts) | _____ | _____ |
| 4) p. 672, #65-69, [strength](18 pts) | _____ | _____ |
| 5) Worksheet-Strength (24 pts) | _____ | _____ |
| 6) p. 647, #12-13, p. 649, #15,-16 [Polyprotic & strength](20 pts) | _____ | _____ |
| 7) p. 653, 24-26; p. 654, #27-29; p. 655, 30-31 [pH & pOH](34pts) | _____ | _____ |
| 8) p.657, #32-34 [Ka](12 pts) | _____ | _____ |
| 9) neutralization worksheet(10 pts) | _____ | _____ |
| 10) p. 664, #44-46; p. 673, #92-93[titration](10 pts) | _____ | _____ |
| 11) p. 665, #47-48(p. 673, #91)[hydrolysis](14 pts) | _____ | _____ |
| 12) Quiz - characteristics of acids , bases & salts | _____ | _____ |
| 13) Quiz - Arrhenius, Bronsted-lowry, Lewis acids, bases, & salts | _____ | _____ |
| 14) Quiz - Neutralization | _____ | _____ |
| 15) Quiz - pH & pOH | _____ | _____ |
| 16) Quiz - Titration | _____ | _____ |

Note Taking Guide: Episode 1101

Arrhenius Definitions: When reacting with _____

- release _____ ions (ex: _____)
 $\text{HCl}_{\text{aq}} \leftrightarrow \text{H}^+ + \text{Cl}^-$
- release _____ ions (ex: _____)
 $\text{NaOH}_{\text{aq}} \leftrightarrow \text{Na}^+ + \text{OH}^-$
- are _____ compounds that release neither _____ ions nor _____ ions.

Solution	Acid (A), Base (B), or Salt (S)	Reacts with Mg (R or NR)	Conductivity (C or NC)
HNO_3	A		
NaOH	B		
NaCl	S		
H_2SO_4			
HCl			
NH_3OH			
KNO_3			
KOH			
$\text{HC}_2\text{H}_5\text{O}_2$			

OPERATIONAL DEFINITIONS:

ACIDS:

- taste _____
- react with some _____ to produce _____ gas
- turn _____ indicators different _____
- react with _____ to produce a _____ and water
- Turns litmus _____

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,2023-05-20 Science Success Class 7 Teacher Resource Book Academic Year 2023-24 *Overcoming Students' Misconceptions in Science* Mageswary Karpudewan, Ahmad Nurulazam, Md Zain, A.L. Chandrasegaran, 2017-02-28 This book discusses the importance of identifying and addressing misconceptions for the successful teaching and learning of science across all levels of science education from elementary school to high school. It suggests teaching approaches based on research data to address students' common misconceptions. Detailed descriptions of how these instructional approaches can be incorporated into teaching and learning science are also included. The science education literature extensively documents the findings of studies about students' misconceptions or alternative conceptions about various science concepts. Furthermore, some of the studies involve systematic approaches to not only creating but also implementing instructional programs to reduce the incidence of these misconceptions among high school science students. These studies however are largely unavailable to classroom practitioners partly because they are usually found in various science education journals that teachers have no time to refer to or are not readily available to them. In response, this book offers an essential and easily accessible guide.

Empowering Science and Mathematics for Global Competitiveness Yuli Rahmawati, Peter Charles Taylor, 2019-06-07 This conference proceedings focuses on enabling science and mathematics practitioners and citizens to

respond to the pressing challenges of global competitiveness and sustainable development by transforming research and teaching of science and mathematics The proceedings consist of 82 papers presented at the Science and Mathematics International Conference SMIC 2018 organised by the Faculty of Mathematics and Natural Sciences Universitas Negeri Jakarta Indonesia The proceedings are organised in four parts Science Science Education Mathematics and Mathematics Education The papers contribute to our understanding of important contemporary issues in science especially nanotechnology materials and environmental science science education in particular environmental sustainability STEM and STEAM education 21st century skills technology education and green chemistry and mathematics and its application in statistics computer science and mathematics education **Teachers' Learning** J. Wallace,W. Louden,2005-12-11 Teachers Learning Stories of Science Education is aimed at science educators who wish for a deeper understanding of how teachers learn to teach science and the role of stories in reporting science education research It is a fascinating look at the knowledge teachers have and use how context influences teachers work and the role of reflection and collaboration in teachers learning At the core of each chapter is a story or group of stories written by or about teachers These stories serve as a form of data to build a set of arguments about how science teachers grow and the possibilities for change in teaching This book is designed for all those involved in the science teaching enterprise Pre service teachers graduate students and science education researchers are invited to utilise both the findings about teachers learning and the research processes employed to develop those findings **First Year Chemistry Students' Conceptions of Acid/base Chemistry** Sally Diane Rupert,2001 **Chapter Resource 2 Chemistry of Life Biology** Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2004 **Biology/science Materials** Carolina Biological Supply Company,1991 **Primary Science Education in East Asia** Yew-Jin Lee,Jason Tan,2018-09-26 This edited volume is a state of the art comparison of primary science education across six East Asian regions namely the People s Republic of China Republic of Korea Republic of China Hong Kong SAR Japan and Singapore While news of educational policies classroom teaching assessment and other educational innovations here often surface in the international media this book brings together for the first time relevant information regarding educational systems and strategies in primary science in East Asia Above all it is a readable yet comprehensive survey readers would have an accurate sense of what has been accomplished what has not worked so well and what remains to be done Invited experts in comparative education research and or science education also provide commentary by discussing common themes across the six regions These types of critical synoptic reviews add much value by enabling readers to understand broad commonalities and help synthesize what must surely be a bewildering amount of very interesting albeit confusing body of facts issues and policies Education in East Asia holds many lessons both positive and negative to offer to the rest of the world to which this volume is a timely contribution to the literature **Educart CBSE Class 12 Chemistry One Shot Question Bank 2026**
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