

acid base titration lab answers

Titration	Acid Formula	Base Formula	Volume of Acid, $V_A$	Molarity of Acid, $M_A$	Volume of Base, $V_B$	End, or Equivalence Point, pH
#1	HCl	NaOH	25.00	0.1353	18.50	7
Reaction Equation	$\text{HCl} + \text{NaOH} \rightarrow \underline{\text{NaCl}} + \underline{\text{H}_2\text{O}}$					
#2	$\text{HNO}_3$	KOH	25.00	0.1644	19.40	7
Reaction Equation	$\text{HNO}_3 + \text{KOH} \rightarrow \underline{\text{KNO}_3} + \underline{\text{H}_2\text{O}}$					

Calculations

Use the equation, Molarity acid x Volume acid = Molarity base x Volume base:  $M_A \cdot V_A = M_B \cdot V_B$

Titration #1:

Titration #2:

$$\text{Molarity of Base, } M_B = \frac{M_A \cdot V_A}{V_B}$$

$$\text{Molarity of Base, } M_A = \frac{M_B \cdot V_B}{V_A}$$

$$\text{Titration \#1: } M_B = \underline{0.1828} \text{ M}$$

$$\text{Titration \#2: } M_A = \underline{0.2119} \text{ M}$$

# Acid Base Titrations Lab Answers

**N Noddings**



## **Acid Base Titrations Lab Answers:**

## Unveiling the Energy of Verbal Artistry: An Psychological Sojourn through **Acid Base Titrations Lab Answers**

In a world inundated with monitors and the cacophony of instantaneous interaction, the profound power and mental resonance of verbal artistry often diminish in to obscurity, eclipsed by the constant barrage of noise and distractions. However, nestled within the musical pages of **Acid Base Titrations Lab Answers**, a fascinating work of fictional splendor that impulses with fresh emotions, lies an remarkable trip waiting to be embarked upon. Penned by a virtuoso wordsmith, that exciting opus books visitors on a mental odyssey, gently revealing the latent possible and profound influence embedded within the delicate internet of language. Within the heart-wrenching expanse of this evocative evaluation, we will embark upon an introspective exploration of the book is main styles, dissect their interesting publishing fashion, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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