

# ACID-BASE TITRATION

Name \_\_\_\_\_

An Acid-Base Titration is the determination of the concentration of an acid or base by exactly neutralizing the acid or base with an acid or base of known concentration.

Example: It takes 45.2 ml. of 0.1 M hydrochloric acid to neutralize 30 ml. of sodium hydroxide. What was the concentration of sodium hydroxide?



Step 1: Express the known concentration in liters .045 L of HCl

Step 2: Multiply by its concentration: (.045 L)(0.1 M) = .0045

Step 3: Multiply by the mole to mole comparison (see equation)  
(.0045) (1/1) = .0045 still

Step 4: Divide by the unknown concentration in liters

$$.0045 / .030 \text{ L} = .150 \text{ M}$$

1. In the titration of 10 ml of vinegar 40 ml of .1 M sodium hydroxide was used. What was the molarity of the vinegar?



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2. If 40 ml. of .1 M sodium hydroxide neutralizes 30 ml. of HCl, what was the molarity of the HCl?



\_\_\_\_\_

3. It took 18 ml. of .2 molarity to titrate 45 ml. of  $\text{Ca}(\text{OH})_2$ . What was the molarity of the  $\text{Ca}(\text{OH})_2$ ?



\_\_\_\_\_

4. 60 ml. of .5 m. of HCl is needed to neutralize 60 ml. of NaOH. Calculate the molarity of the NaOH



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5. To titrate 30 ml. of potassium hydroxide required 20 ml. of .1 M Hydrochloric Acid. Write the equation and tell the molarity of the potassium hydroxide.

\_\_\_\_\_

# Acid Base Titration Worksheet Answers

**Adam Hulanicki**



## Acid Base Titration Worksheet Answers:

**First Year Chemistry Students' Conceptions of Acid/base Chemistry** Sally Diane Rupert, 2001      **Teaching Science Online** Dietmar Kennepohl, 2023-07-03 With the increasing focus on science education growing attention is being paid to how science is taught Educators in science and science related disciplines are recognizing that distance delivery opens up new opportunities for delivering information providing interactivity collaborative opportunities and feedback as well as for increasing access for students This book presents the guidance of expert science educators from the US and from around the globe They describe key concepts delivery modes and emerging technologies and offer models of practice The book places particular emphasis on experimentation lab and field work as they are fundamentally part of the education in most scientific disciplines Chapters include Discipline methodology and teaching strategies in the specific areas of physics biology chemistry and earth sciences An overview of the important and appropriate learning technologies ICTs for each major science Best practices for establishing and maintaining a successful course online Insights and tips for handling practical components like laboratories and field work Coverage of breaking topics including MOOCs learning analytics open educational resources and m learning Strategies for engaging your students online      **Experimental Chemistry** Robert J. Artz, 1982      **Educart ICSE Class 10 One-shot Question Bank 2026 Chemistry (strictly for 2025-26 boards)** Sir Tarun Rupani, 2025-07-12 Fast track your Chemistry revision with this exam ready resource This One shot Question Bank by Sir Tarun Rupani is designed to help ICSE Class 10 students revise the complete Chemistry syllabus quickly and thoroughly It simplifies theory boosts numerical accuracy and ensures strong exam practice all aligned with the 2025 26 ICSE syllabus Key Features Strictly Based on ICSE 2025 26 Curriculum Complete chapter coverage including Periodic Table Chemical Bonding Acid Base Organic Chemistry and more One shot Format Each chapter includes concise concept notes chemical equations reactions and key diagrams for quick recall Complete Coverage of Question Types Includes objective short long answers equation based numerical and reasoning questions Chapterwise PYQs Included Practice with previous years ICSE board questions to understand trends and improve retention Solved Answers in ICSE Format Clear well structured solutions using proper units chemical symbols and balanced equations Smart Revision Focus Special tips to avoid common mistakes in writing reactions balancing equations and attempting numericals Why Choose This Book This Chemistry One shot by Sir Tarun Rupani is built for smart preparation whether you re revising at the last minute or practising throughout the term It helps you approach each question with clarity confidence and the precision needed to score high in the 2026 ICSE board exam      **ACID-BASE TITRATION** NARAYAN CHANGDER, 2024-04-01 Note Anyone can request the PDF version of this practice set workbook by emailing me at cbsenet4u gmail com You can also get full PDF books in quiz format on our youtube channel <https://www.youtube.com/smartquiziz> I will send you a PDF version of this workbook This book has been designed for candidates preparing for various competitive examinations It contains many objective questions specifically designed for

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Acid-base Titrations in Nonaqueous Solvents James Sherwood Fritz, 1952      Aqueous Acid-base Equilibria and Titrations Robert De Levie, 1999 This book will give students a thorough grounding in pH and associated equilibria material absolutely fundamental to the understanding of many aspects of chemistry It is in addition a fresh and modern approach to a topic all too often taught in an outmoded way This book uses new theoretical developments which have led to more generalized approaches to equilibrium problems these approaches are often simpler than the approximations which they replace Acid base problems are readily addressed in terms of the proton condition a convenient amalgam of the mass and charge constraints of the chemical system considered The graphical approach of Bjerrum Hagg and Sillen is used to illustrate the orders of magnitude of the concentrations of the various species involved in chemical equilibria Based on these concentrations the proton condition can usually be simplified often leading directly to the value of the pH In the description of acid base titrations a general master equation is developed It provides a continuous and complete description of the entire titration curve which can then be used for computer based comparison with experimental data Graphical estimates of the steepness of titration curves are also developed from which the practicality of a given titration can be anticipated Activity effects are described in detail including their effect on titration curves The discussion emphasizes the distinction between equilibrium constants and electrometric pH measurements which are subject to activity corrections and balance equations and spectroscopic pH measurements which are not Finally an entire chapter is devoted to what the pH meter measures and to the experimental and theoretical uncertainties involved

**General Theory of Acid-base Titration** Pat Richard Pondy, 1949      *Working with Chem Separate* Wink, 1999-12-01      **Acid Base Titrations** Shah Purvesh, 2015-08-04 In Acid Base Titrations the author discussed various acid base titration It gives some basic concept of acid base titration The analysis of acid base titration discussed in this book It also covers Titration of polyprotic acid and mixture of acids Titration of a Polybasic base with a Strong Acid pH transition range for an indicator titration of salts Differential Alkali titration are discussed in simplest type

**General Expressions for Acid-Base Titrations of Arbitrary Mixtures**, 1995 A single

general master equation is given for acid base titrations describing the entire progress of the titration and equally valid for the titration of a strong acid with a strong base as for that of the titration of an arbitrary mixture of acids with an arbitrary mixture of bases or vice versa     Acid-base Chemistry Clifford W. Hand, Harry Lyon Blewitt, 1986 Very Good No Highlights or Markup all pages are intact     *Acid-Base Titration* Bird, 1995-01-01     *Acid-Base Titration in Non-Aqueous Solvents* James S. Fritz, 1973-06-01     *Determination of acetic acid in vinegar using acid-base titration* Jelka Sodja-Božič, Tomaž Ogrin, 2008     **Reactions of Acids and Bases in Analytical Chemistry** Adam Hulanicki, 1987     Weak Acid - Strong Base Titration , 200?     **ACID-BASE EQUILIBRIA** NARAYAN CHANGDER, 2024-05-16 Note Anyone can request the PDF version of this practice set workbook by emailing me at cbsenet4u gmail com You can also get full PDF books in quiz format on our youtube channel <https://www.youtube.com/smartquizz> I will send you a PDF version of this workbook This book has been designed for candidates preparing for various competitive examinations It contains many objective questions specifically designed for different exams Answer keys are provided at the end of each page It will undoubtedly serve as the best preparation material for aspirants This book is an engaging quiz eBook for all and offers something for everyone This book will satisfy the curiosity of most students while also challenging their trivia skills and introducing them to new information Use this invaluable book to test your subject matter expertise Multiple choice exams are a common assessment method that all prospective candidates must be familiar with in today s academic environment Although the majority of students are accustomed to this MCQ format many are not well versed in it To achieve success in MCQ tests quizzes and trivia challenges one requires test taking techniques and skills in addition to subject knowledge It also provides you with the skills and information you need to achieve a good score in challenging tests or competitive examinations Whether you have studied the subject on your own read for pleasure or completed coursework it will assess your knowledge and prepare you for competitive exams quizzes trivia and more     **Acid-Base Diagrams** Heike Kahlert, Fritz Scholz, 2013-07-31 Understanding acid base equilibria made easy for students in chemistry biochemistry biology environmental and earth sciences Solving chemical problems be it in education or in real life often requires the understanding of the acid base equilibria behind them Based on many years of teaching experience Heike Kahlert and Fritz Scholz present a powerful tool to meet such challenges They provide a simple guide to the fundamentals and applications of acid base diagrams avoiding complex mathematics This textbook is richly illustrated and has full color throughout It offers learning features such as boxed results and a collection of formulae     **Acid-bases in Analytical Chemistry** Izaak Maurits Kolthoff, 1966

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

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