19.1 Acid-Base Theories [587]: Section Assessment

- Due Apr 23 at 11:59pm
- Points 16
- Questions 16
- Available Apr 18 at 12am May 24 at 11:59pm
- Time Limit None
- Allowed Attempts 3

Take the Quiz Again

Attempt History

	Attempt	Time	Score	
LATEST	Attempt 1	1,426 minutes	14 out of 16	

① Answers will be shown after your last attempt

Score for this attempt: 14 out of 16

Submitted Apr 19 at 4:19pm

This attempt took 1,426 minutes.

- 88

Question 1

1 / 1 pts

Write a chemical equation for the ionization of HNO₃ in water.

- A) $HNO_3 + H_2O \rightarrow H_3O^+ + NO_3^-$
- B) $HNO_3 + H_2O \rightarrow H_3O + NO_3$
- C) $HNO_3 + H_2O \rightarrow H_3O + NO_3$
- D) $HNO_3 + H_2O \rightarrow OH^- + H_2NO_3^+$
- @ D
- @ C
- B
- (B) A

22

Question 2

1 / 1 pts

Write a chemical equation for the reaction of CO₃2- with water.

Acid Base Theories Section Review Answers Chemistry

T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder

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